

Alkene Ozonolysis: Various Transformation of Ozonide

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College of Chemistry and Molecular Engineering

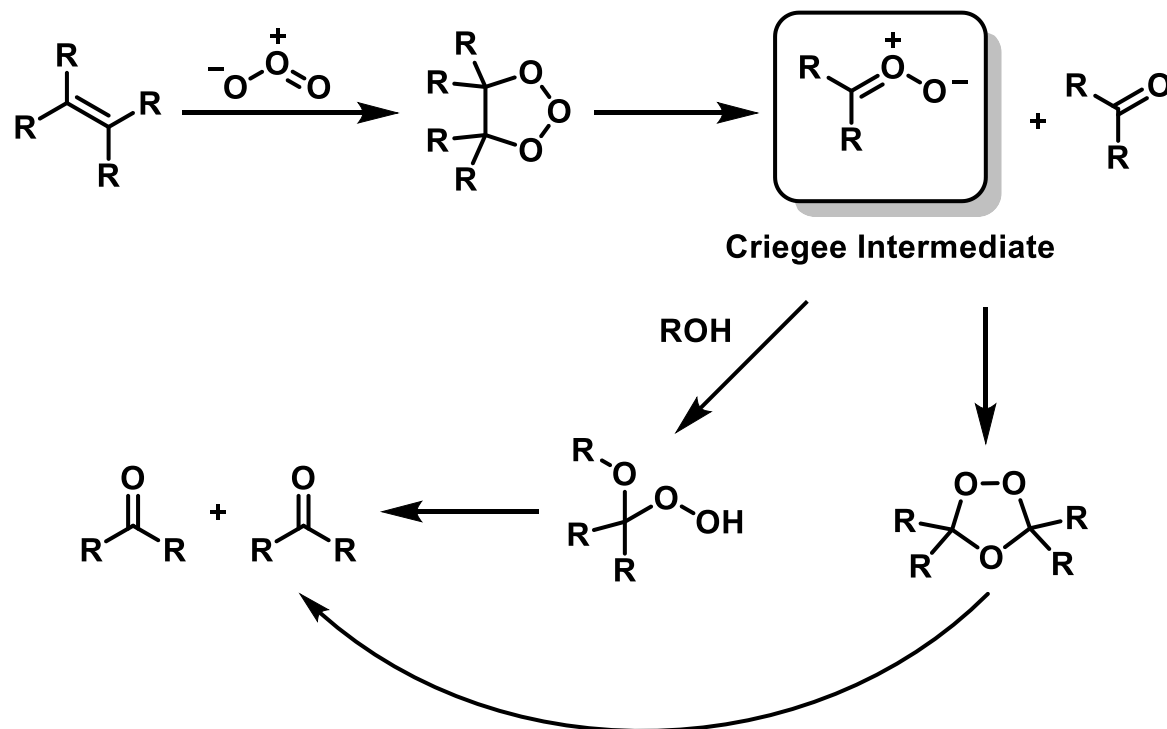
December 16th 2023

Outline

- **Introduction & Earlier Capture Reactions**
- **Criegee Intermediates: Generation & Transformation**
- **Transformation of Secondary Ozonides**
- **Photoexcited Nitroarenes**
- **Summary**

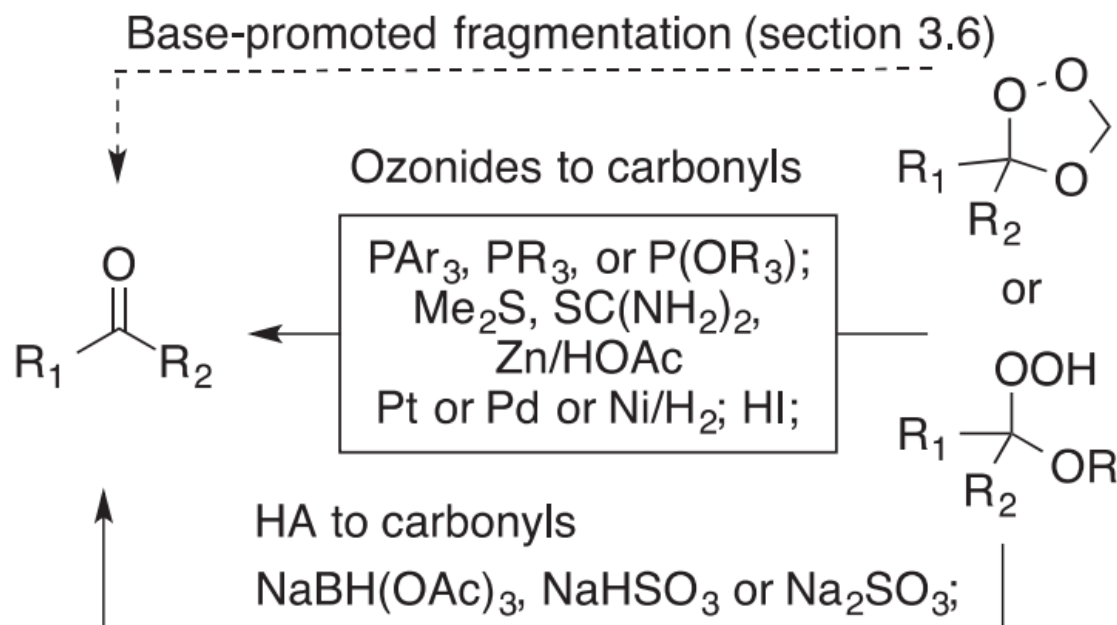
Introduction

- Ozone: a very electrophilic 1,3-dipole
- Alkene: More reactive than alkyne or C-H
- General & Selective method of cleaving double bonds
- **A Reservoir of High Oxidation State**

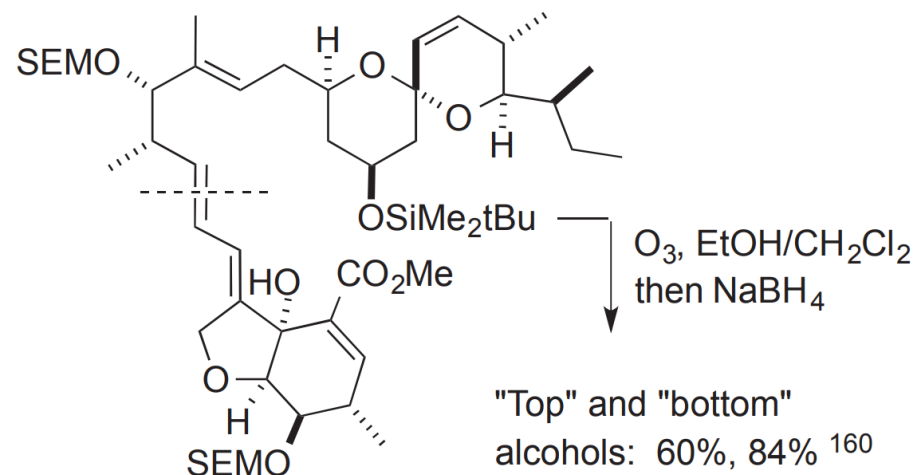
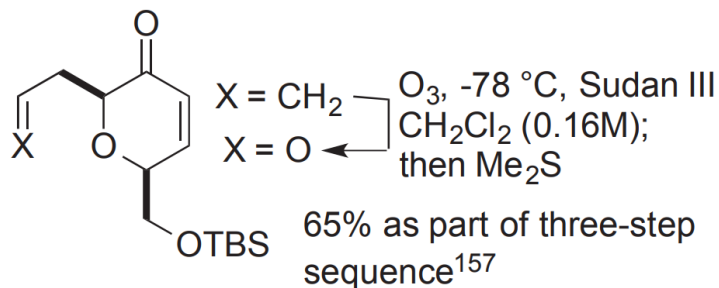
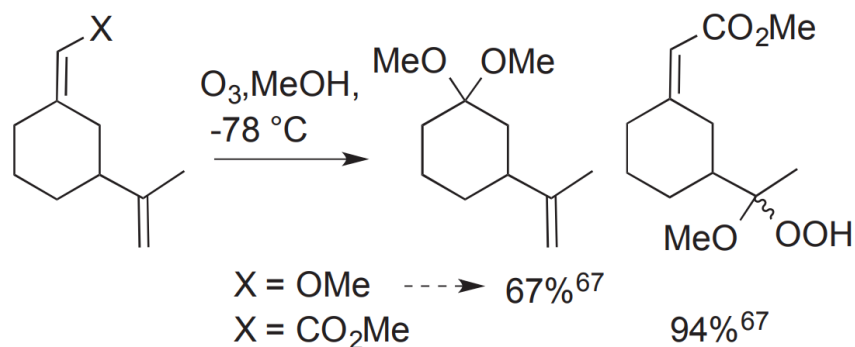
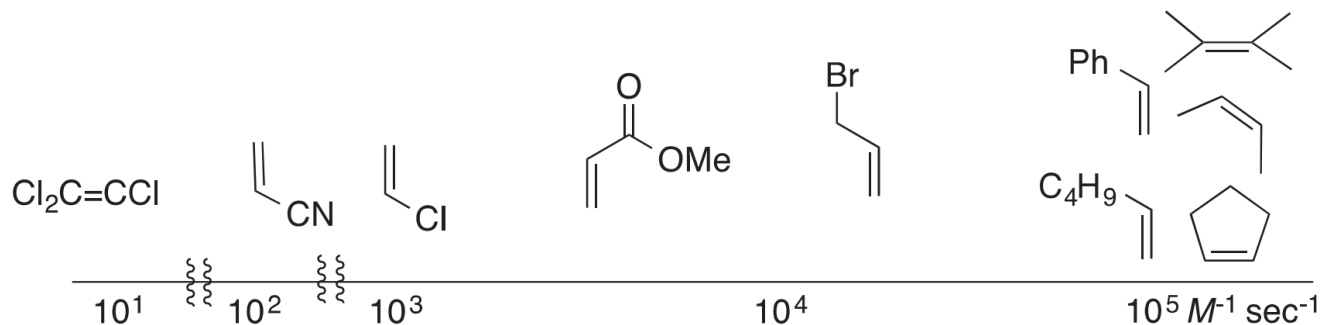


Introduction

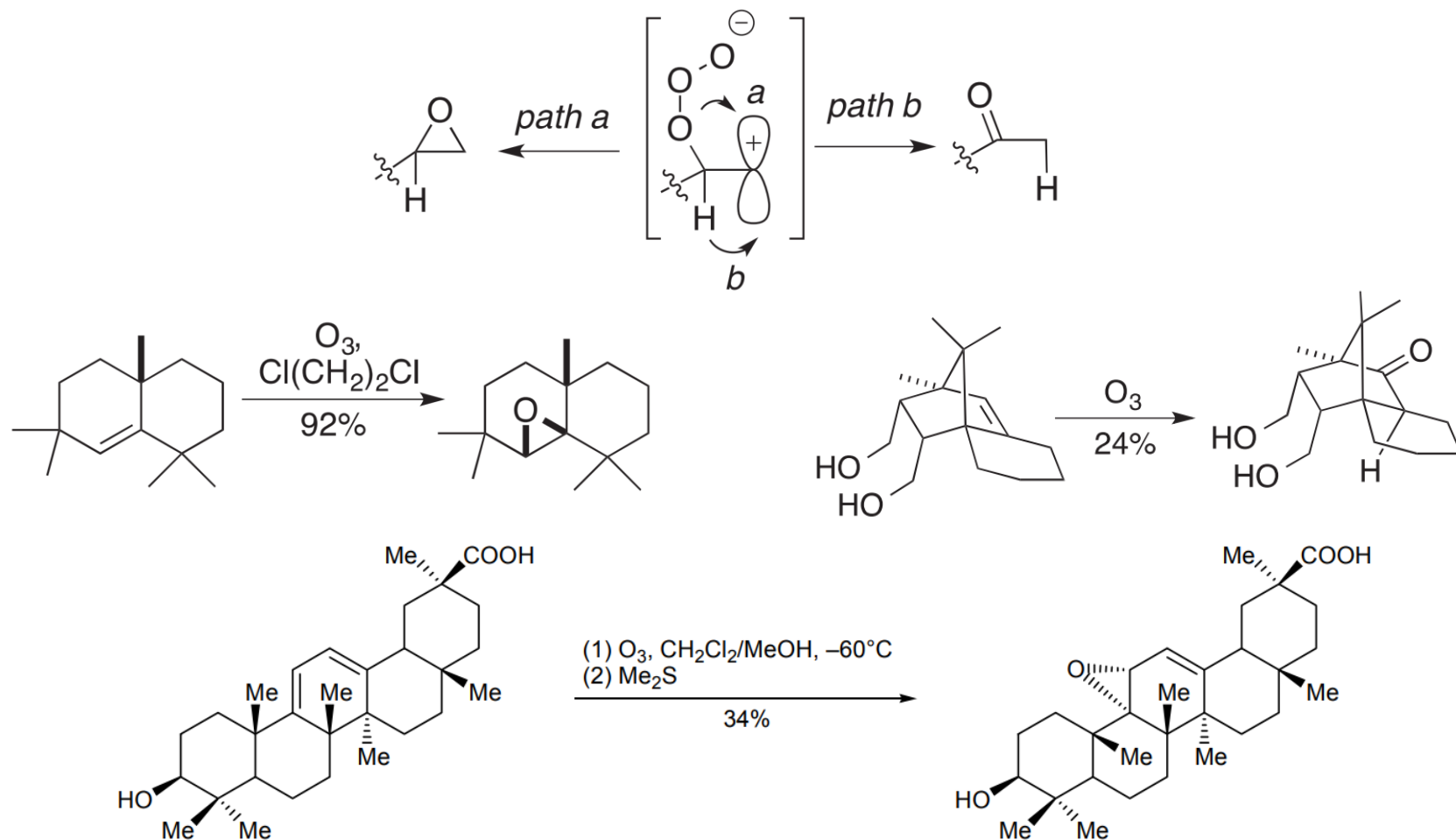
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- Alkene: More reactive than alkyne or C-H
- General & Selective method of cleaving double bonds
- **A Reservoir of High Oxidation State**



Regioselectivity about Alkenes

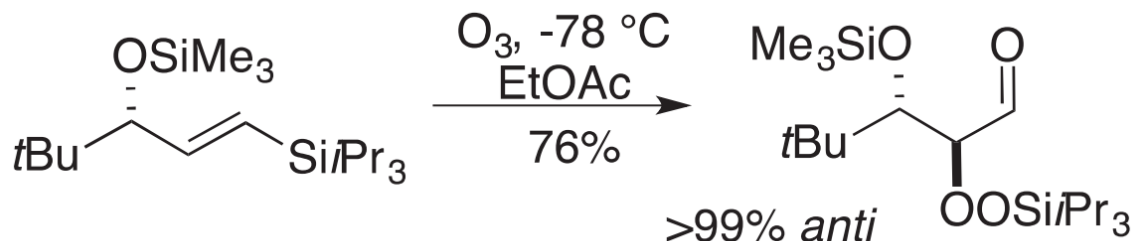
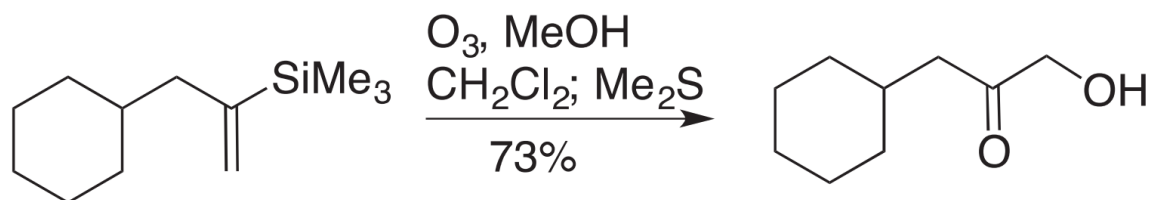
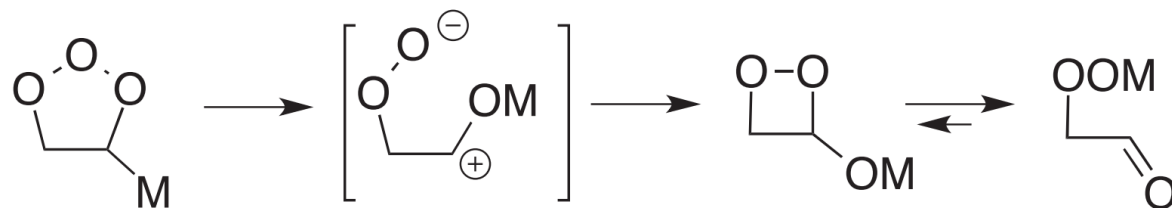


Alkenes with Large Steric Hindrance

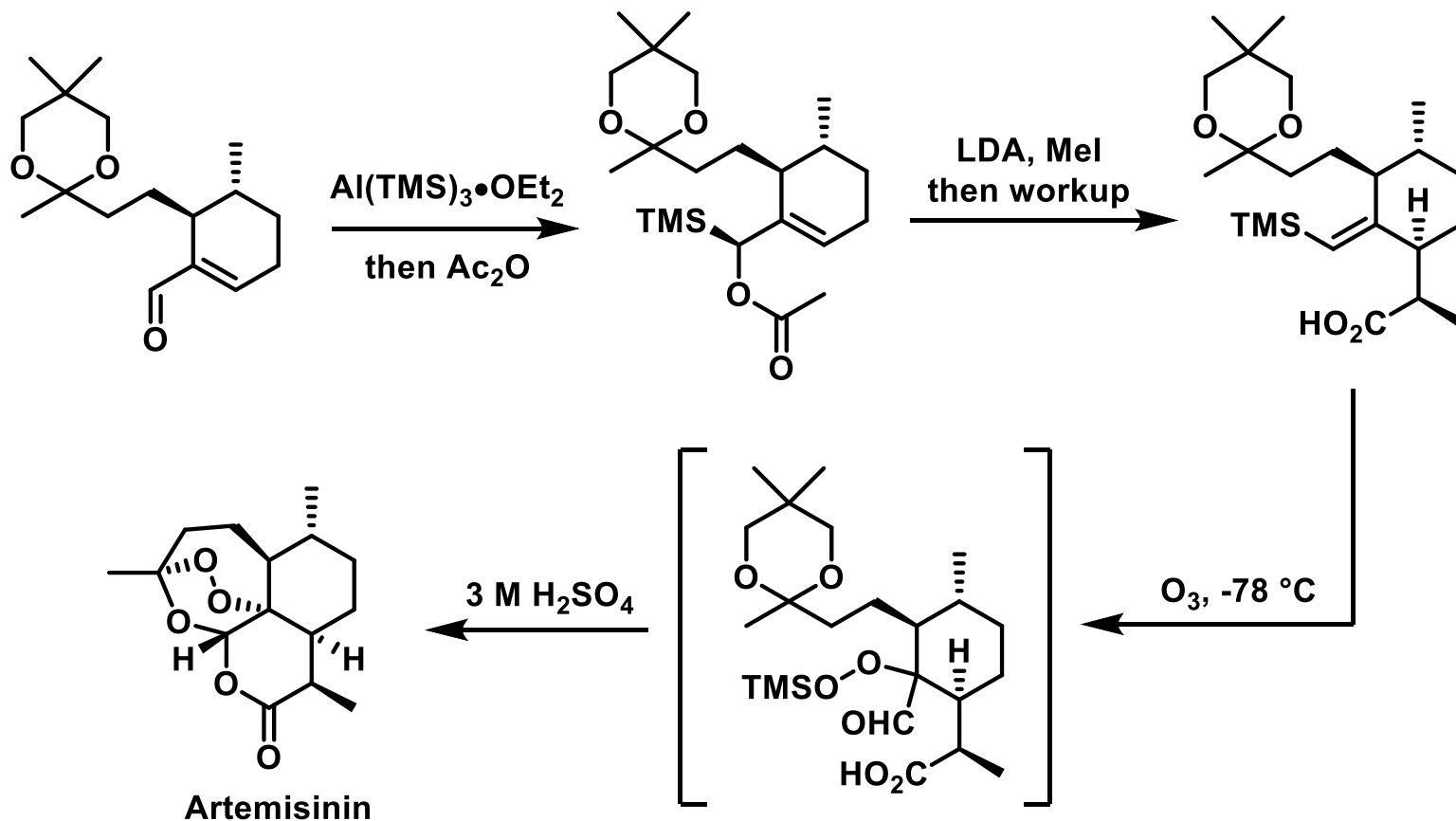


Budaev, A. et al. *Chem. Nat. Compd.* **2014**, 50, 302; Hochstetler, A. *J. Org. Chem.* **1975**, 40, 1536;
Kondolff, I. et al. *Tetrahedron* **2007**, 63, 9100.

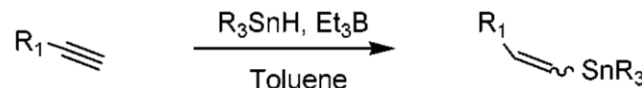
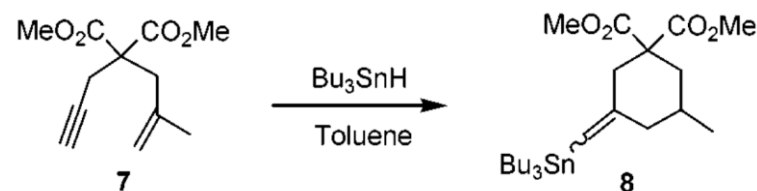
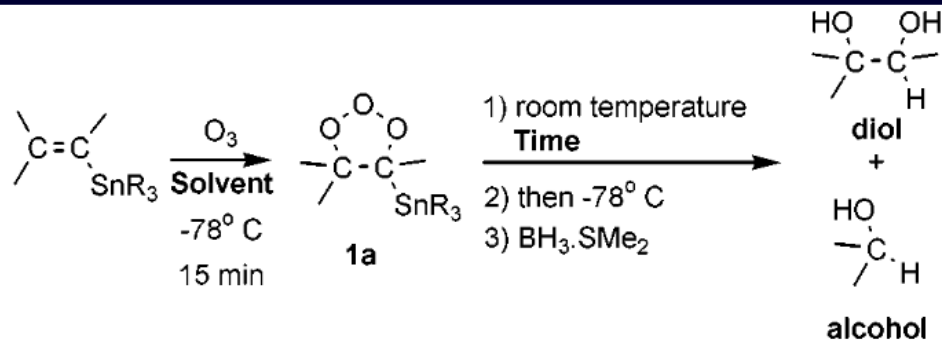
Interrupted at Primary Ozonide



Synthesis of Artemisinin



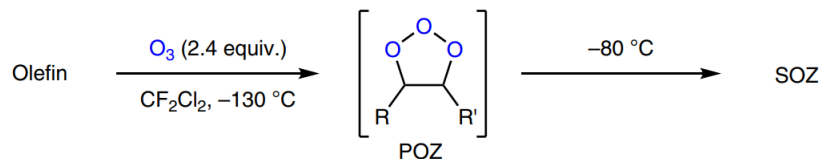
Unusual Stable Primary Ozonide: Tin

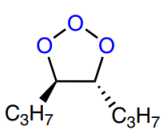
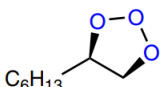


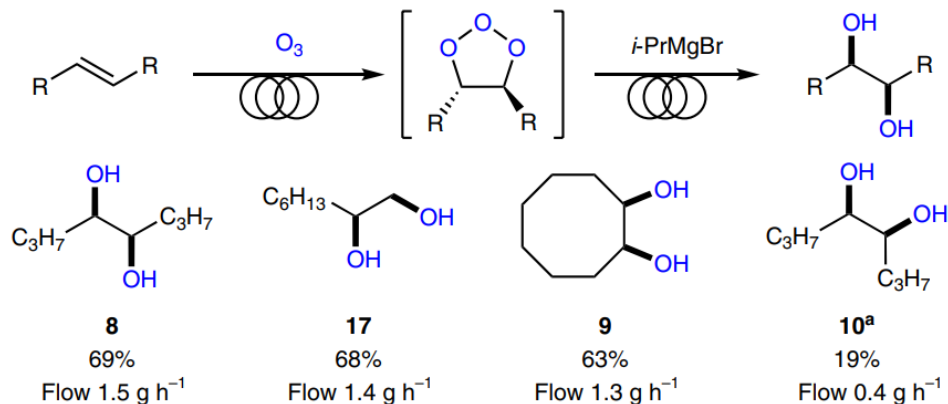
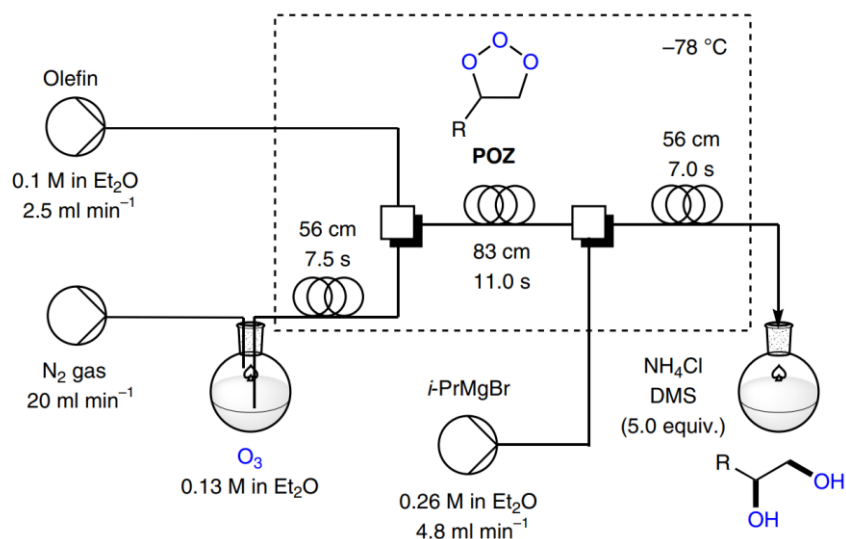
entry	substrate	solvent	time (h)	ratio diol:alcohol	yield (%)
i	8	MeOH	0 ^a	1:0	67
ii	8	MeOH	0.5	1:2	62
iii	8	MeOH	2	1:8	63
iv	8	MeOH	4	0:1	65
v	8	EtOAc	0 ^a	1:0	59
vi	8	EtOAc	2	1:1	63
vii	8	CH_2Cl_2	0 ^a	1:0	65
viii	8	CH_2Cl_2	2	1:2	61
ix	10a	MeOH	0 ^a	5:1	71
x	10a	MeOH	2	1:3	68
xi	10a	EtOAc	0 ^a	2:1	69
xii	10a	EtOAc	2	1:5	66
xiii	10a	CH_2Cl_2	0 ^a	4:1	62
xiv	10a	CH_2Cl_2	2	1:7	58
xv	10a	toluene	0 ^a	1:1.5	59



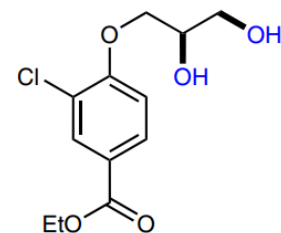
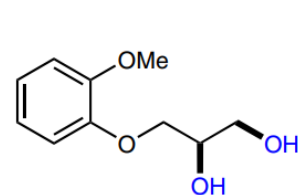
Capture of POZ: Continuous Flow



Entry	Structure	$k_{\text{obs}} \times 10^{-3} (\text{s}^{-1})$	k_{rel}	$t_{1/2} (\text{min})$
POZ 1		1.68 ± 0.02	1.0	6.8 ± 0.2
POZ 2		2.49 ± 0.05	1.48	4.6 ± 0.1



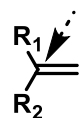
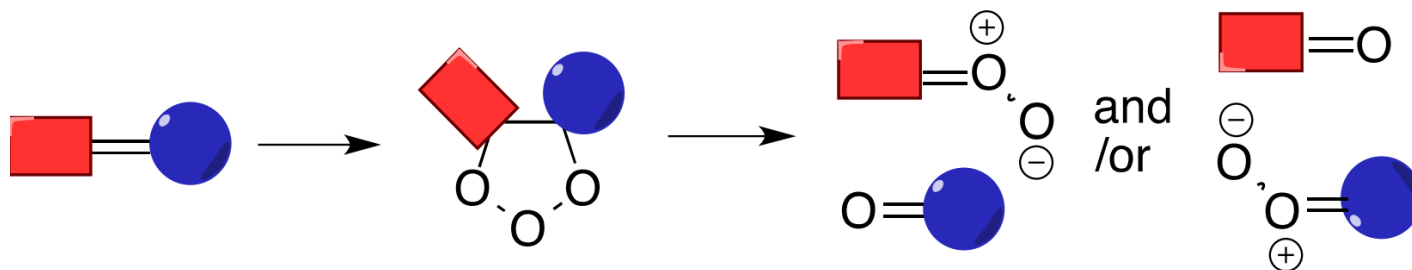
Pharmaceuticals



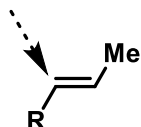
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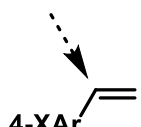
Generated from POZ: Selectivity?



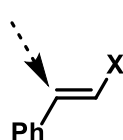
R_1	R_2	
Me	H	62%
Me	Ph	93%
Ph	H	60%
Cy	H	86%



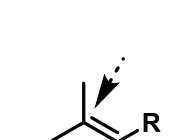
R	
tBu	88%
iPr	71%
Ph	82%



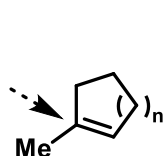
X	
H	60%
Me	71%
OMe	79%
NO_2	14%



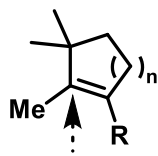
X	
CH_2OH	74%
CH_2Cl	83%
CO_2Me	77%
COMe	10%



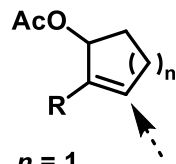
R	
Me	77%
Ph	95%



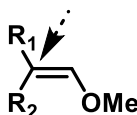
$n = 1$: 95%
 $n = 2$: 100%



$n = 1$
 $R = H$: 95%
 $n = 2$
 $R = H$: 100%
 $R = Me$: 0%



$n = 1$
 $R = H$: 98%
 $R = Me$: 81%
 $n = 2$
 $R = H$: 100%
 $R = Me$: 67%



R_1	R_2	
H	H	95%
Ph	H	95%
Hep	Ph	93%
$-(CH_2)_5-$		80%

EDG: Resonance

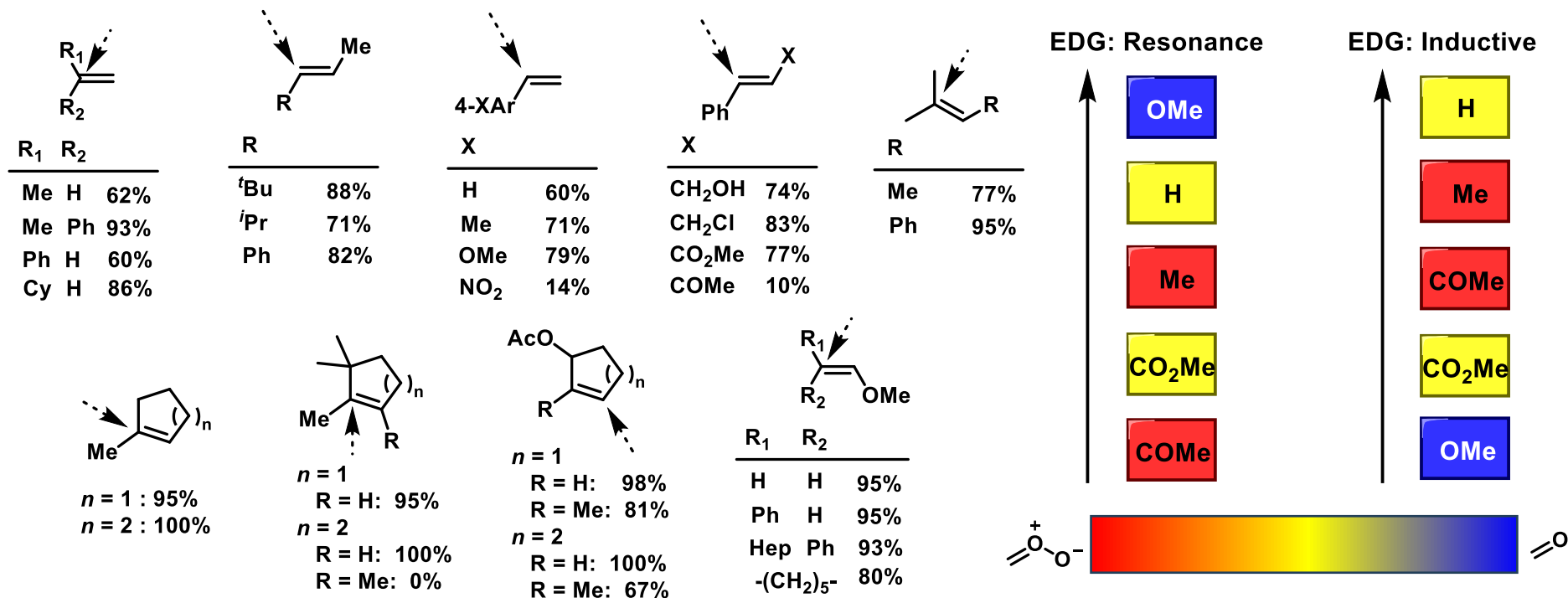


EDG: Inductive

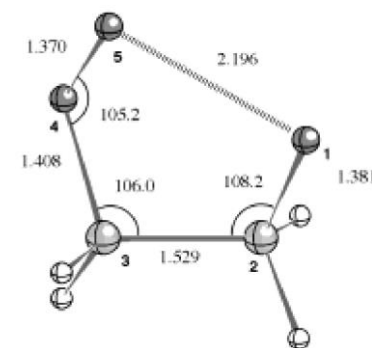
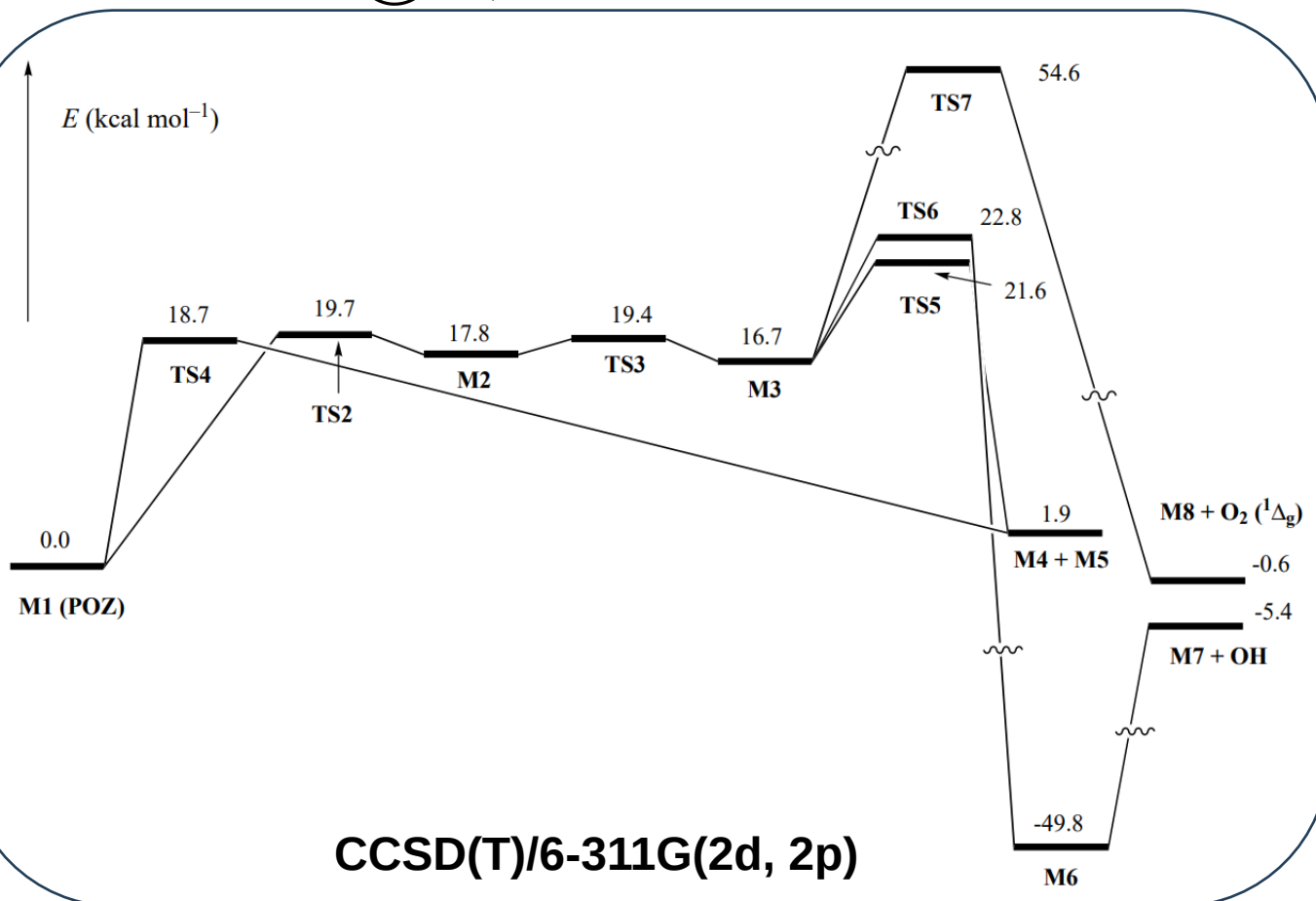
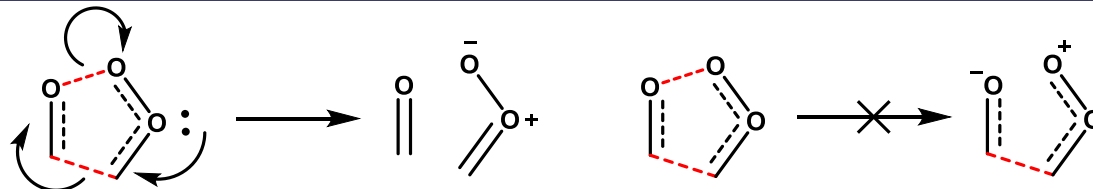


Generated from POZ: Selectivity?

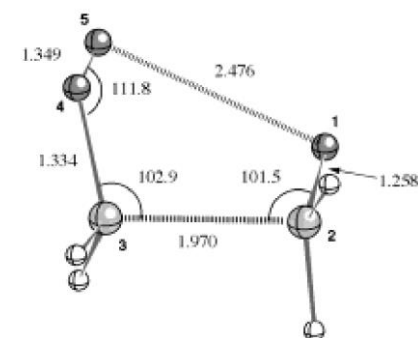
Criegee Tendency:



Which Bond **Heterolysis** First?

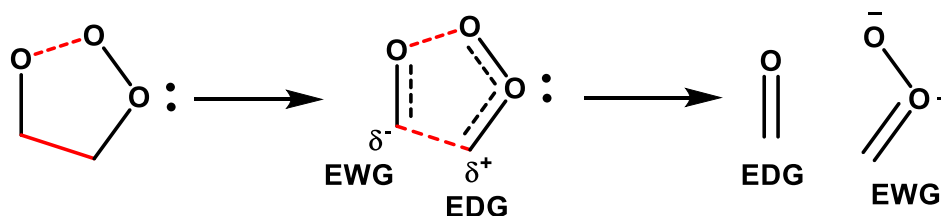


TS2

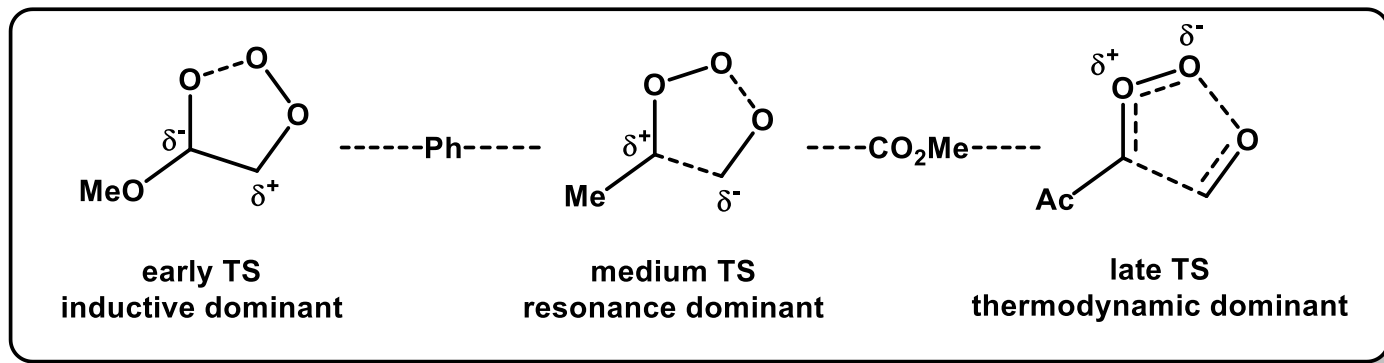


TS4

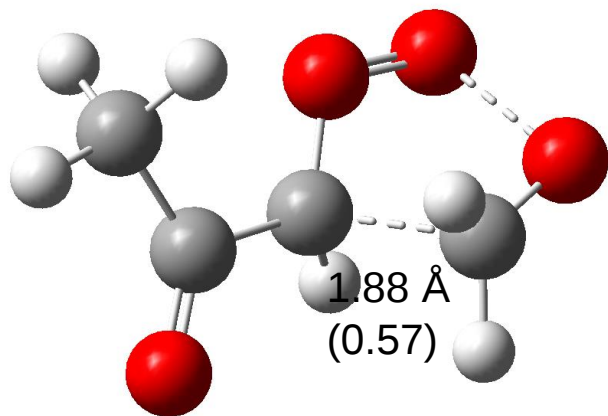
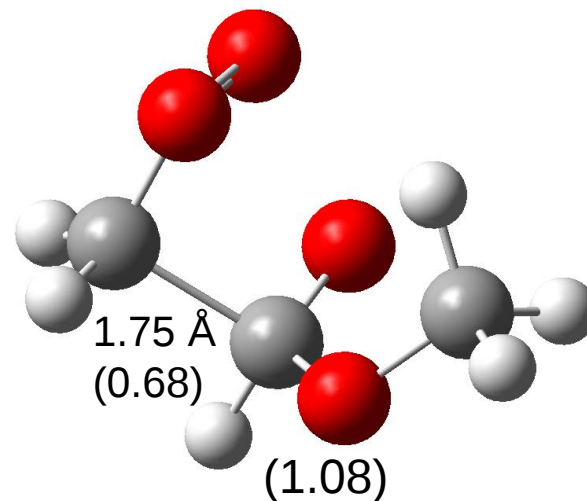
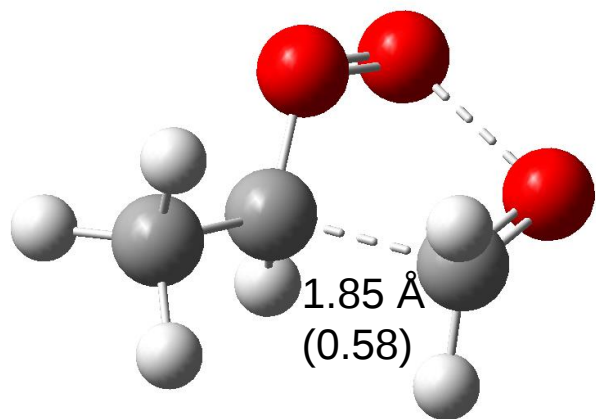
A Proposal: Reaching TS at What Point?



- C-C bond polarization: Criegee to Carbonyl
- Electron-rich degree: Criegee > Carbonyl
- Determining factor: **Early** or **Late TS**?
- **EDGs** cause **Negative ΔH** & **Early TS**



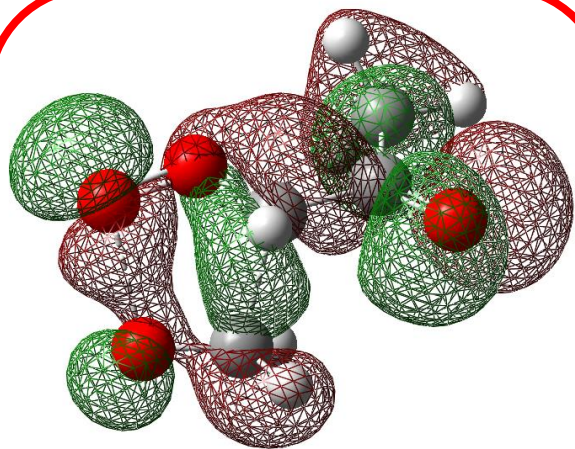
Computation: What is the Critical Point?



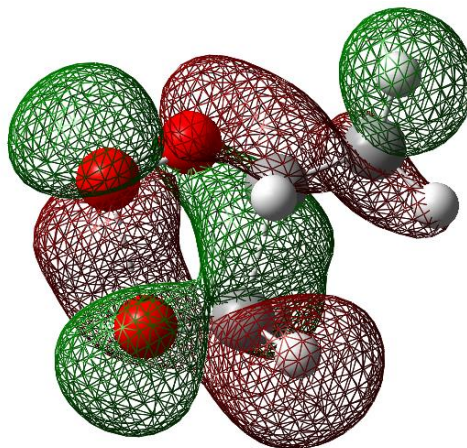
- Me vs Ac: Similar bond length & order in TS
- Only dimension for **Early** or **Late**?
- Orbital Interaction
- **Microscopic Reversibility**

M06-2X(D3)/def2SVP SMD = Methanol

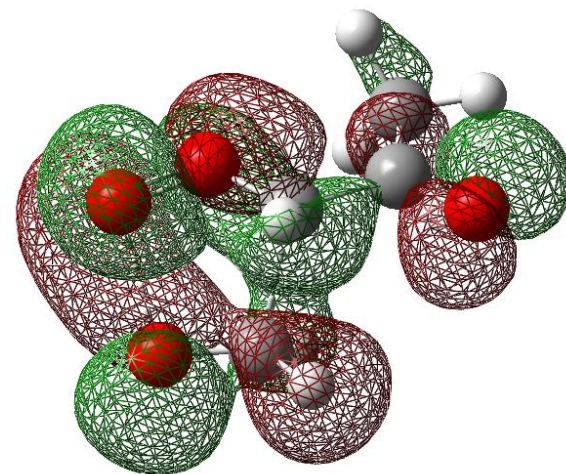
Product-Like Orbital Interaction?



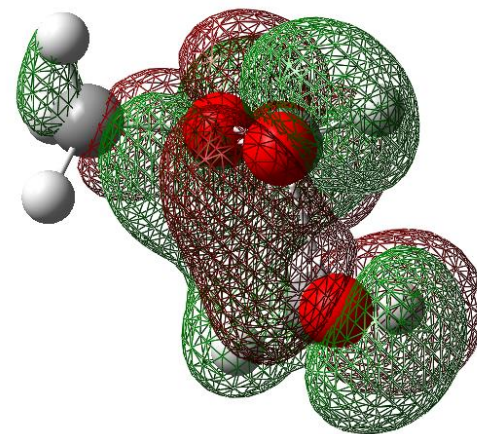
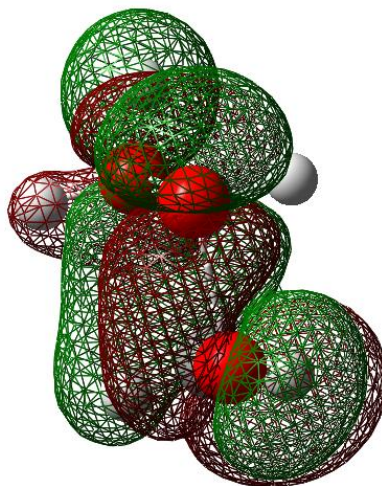
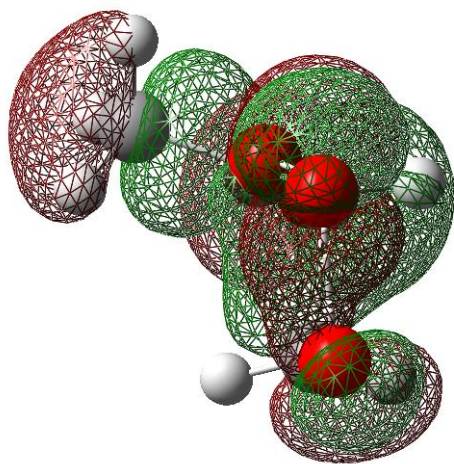
HOMO-1 (Ac)



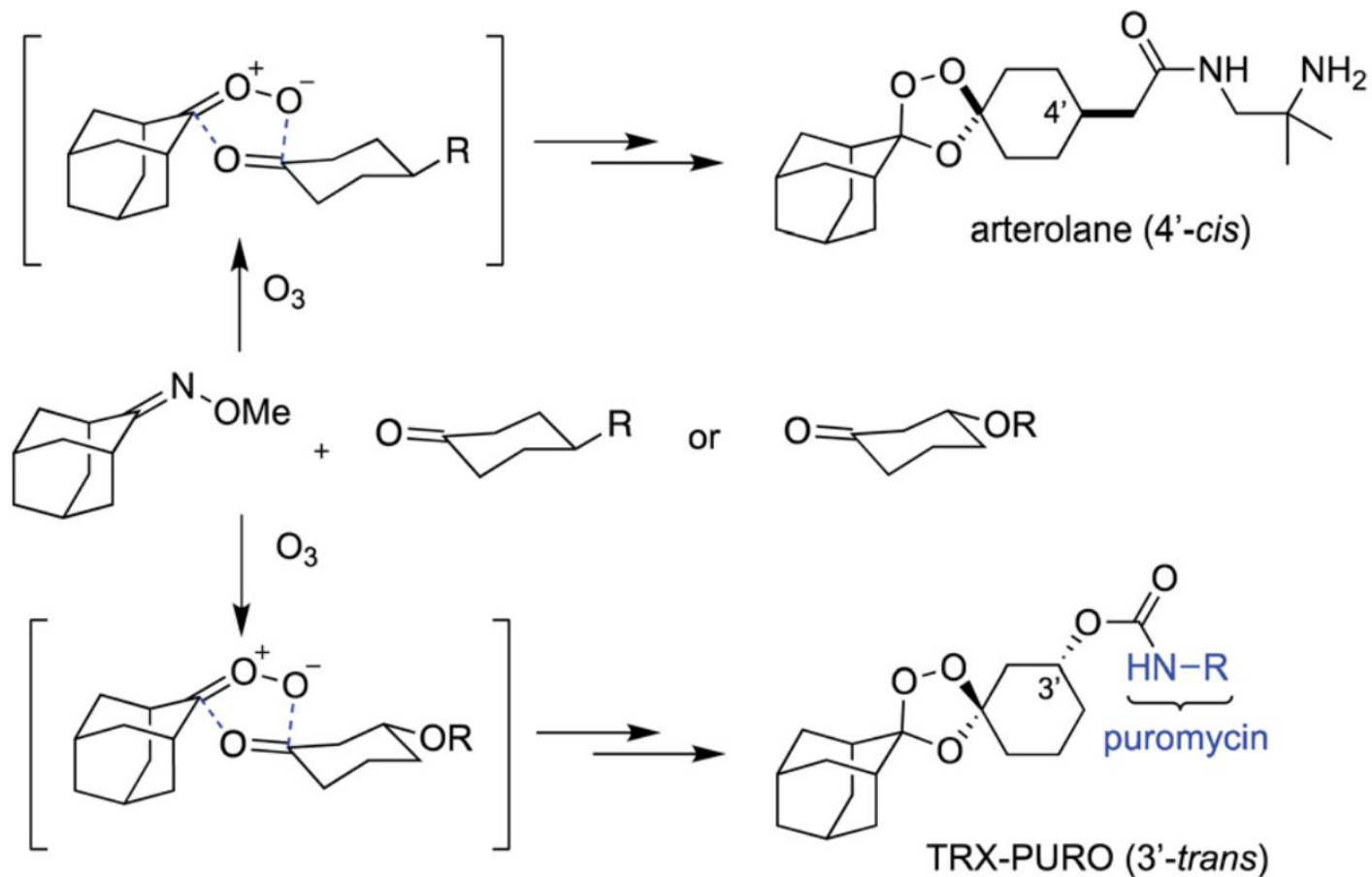
HOMO-1 (Me)



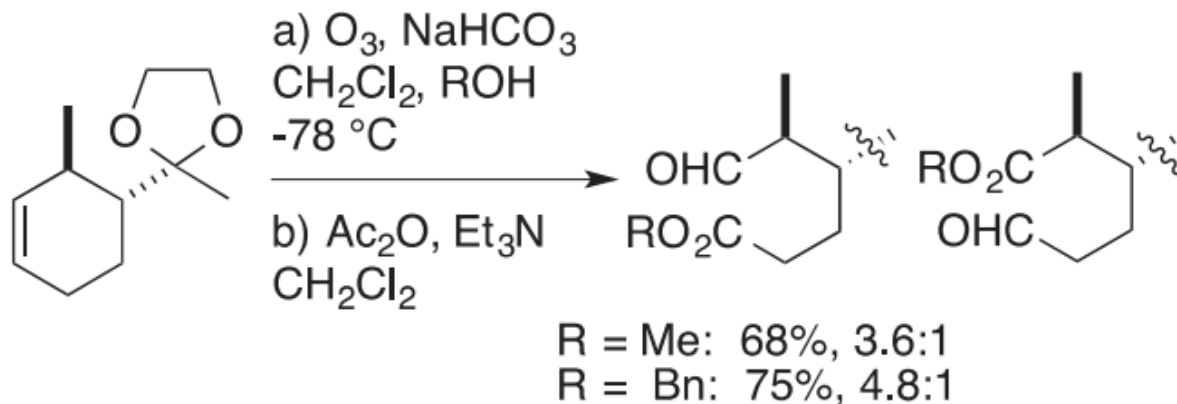
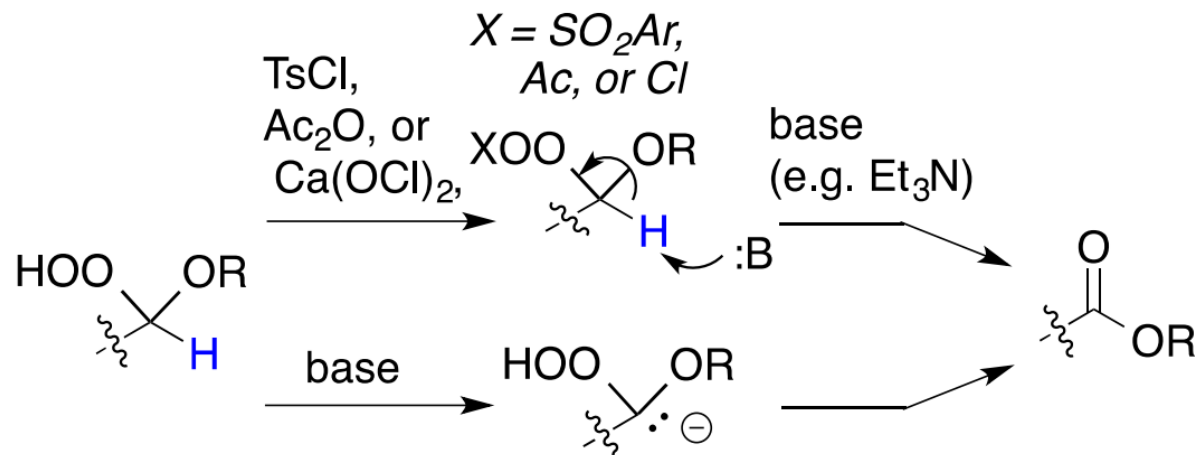
HOMO-2 (Ac)



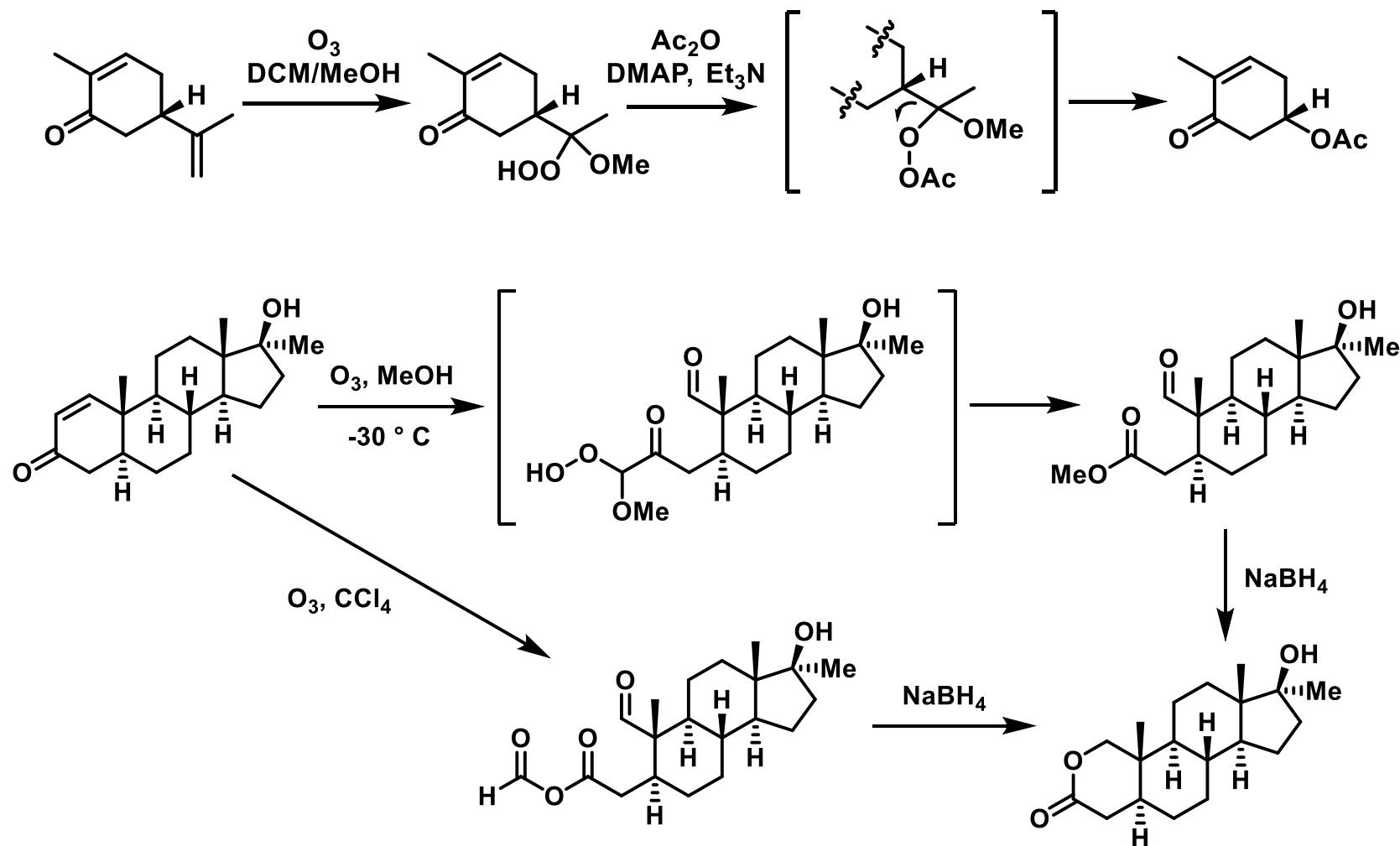
Griesbaum Ozonolysis



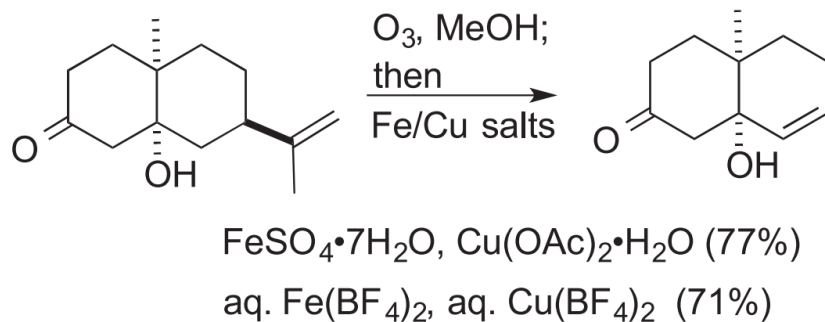
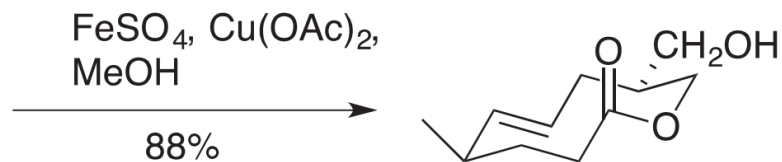
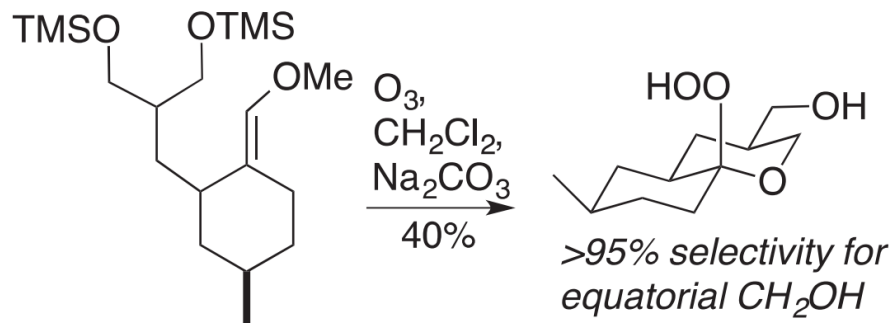
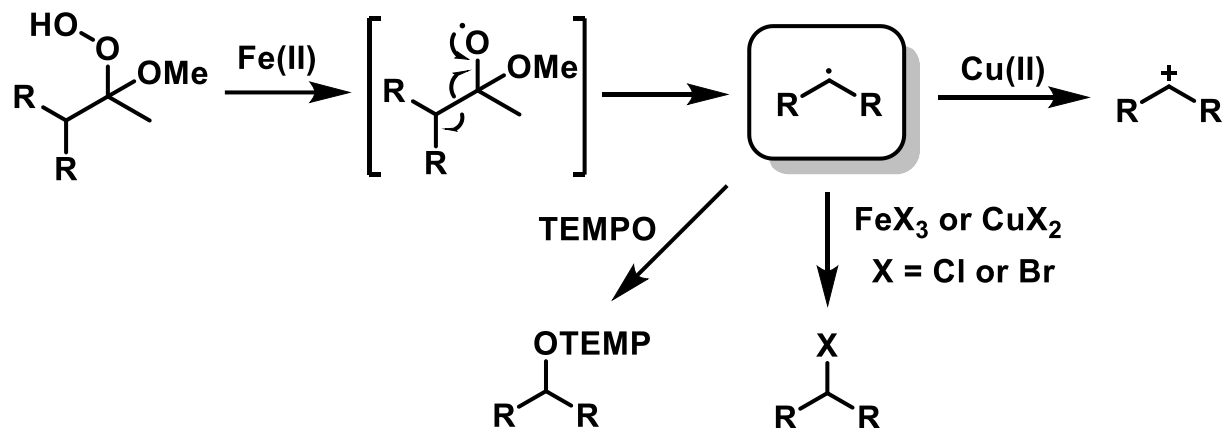
Hydroperoxyl Acetal to Ester



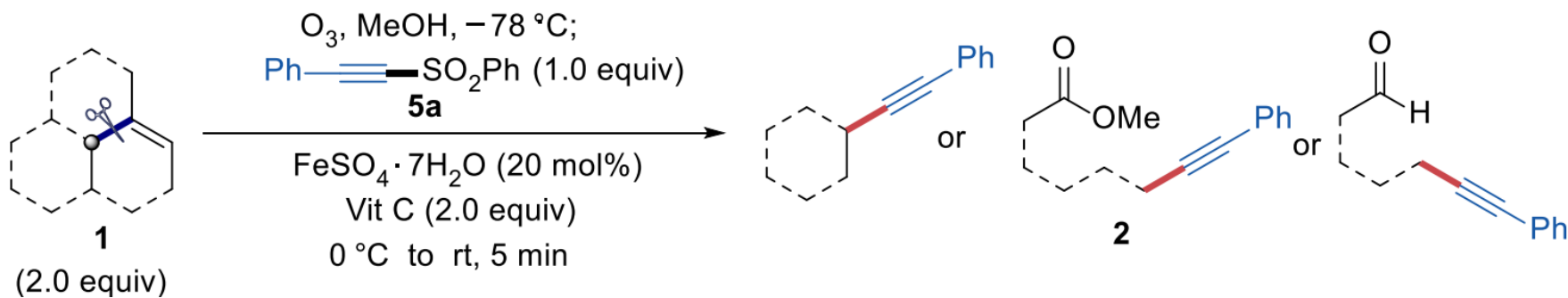
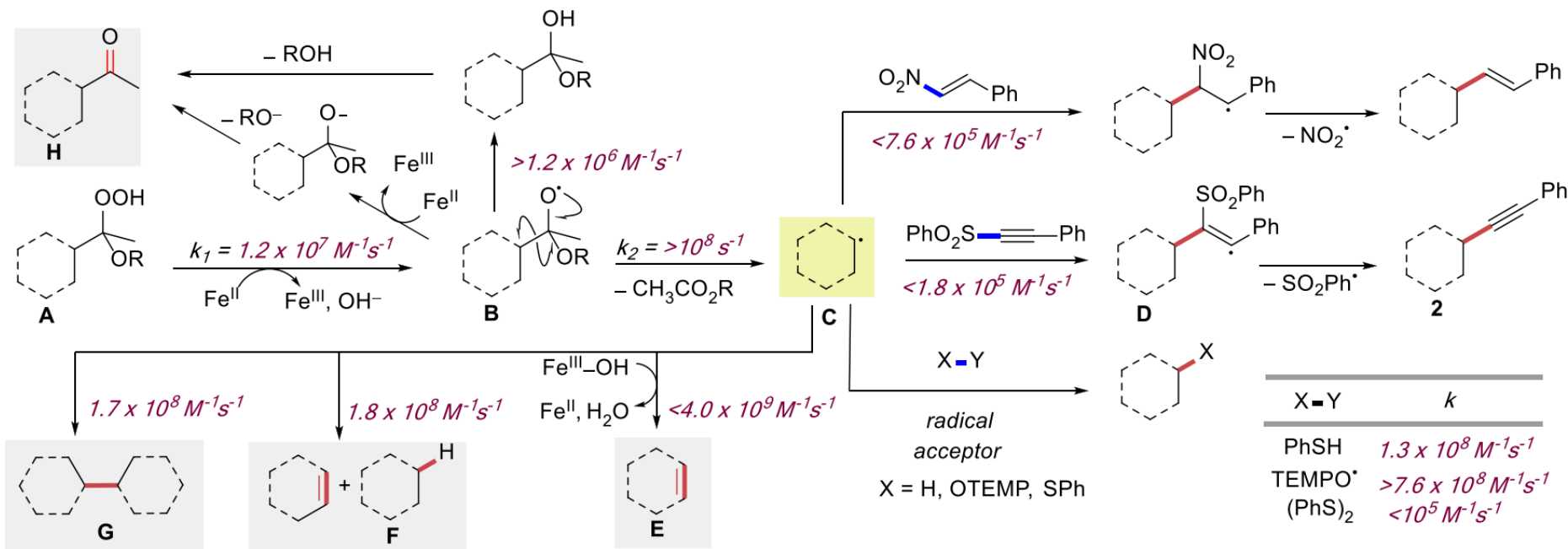
Formal Baeyer-Villiger Oxidation



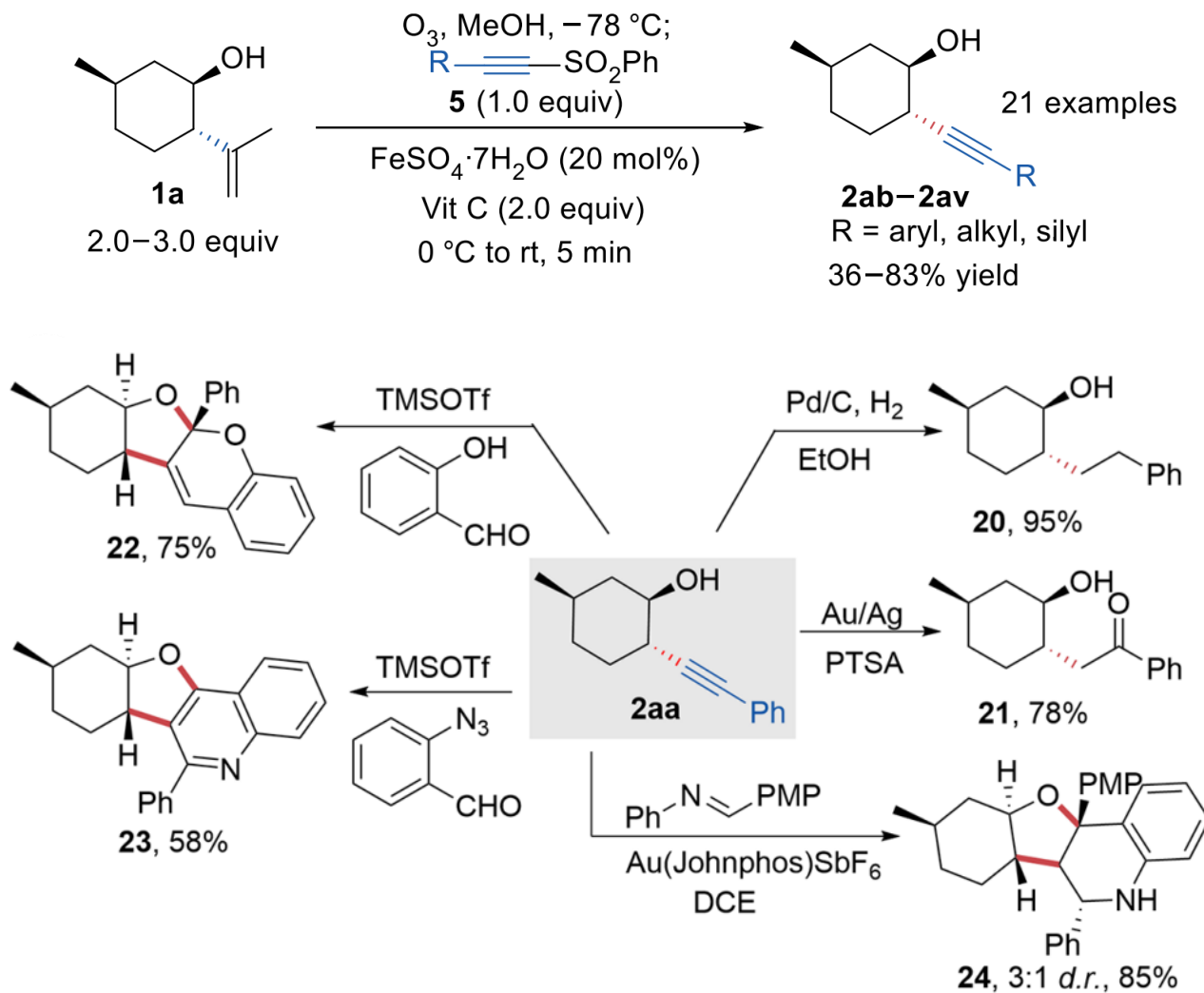
As Radical Precursor



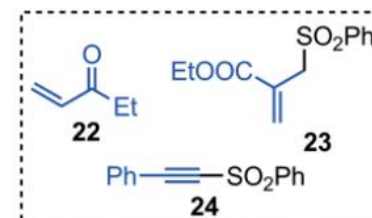
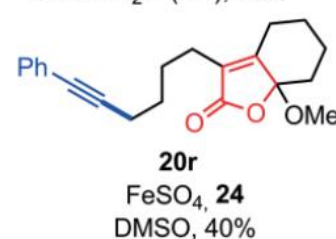
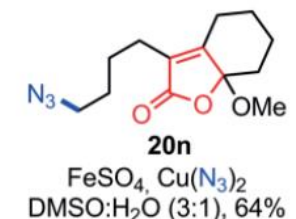
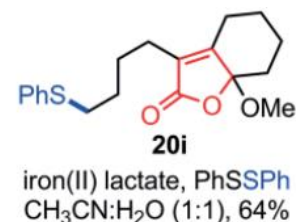
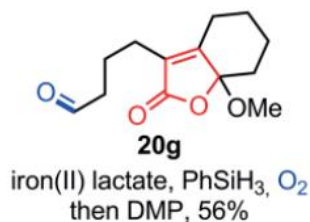
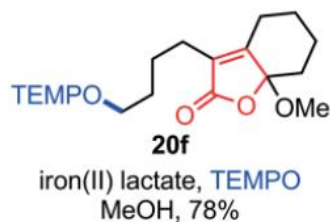
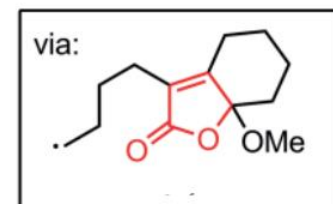
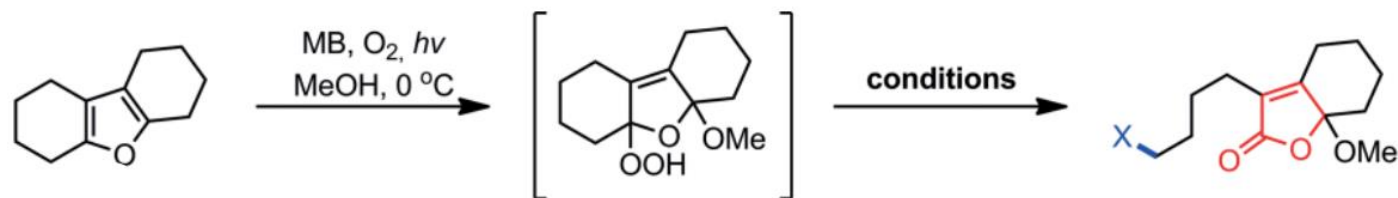
Radical Addition Reaction?



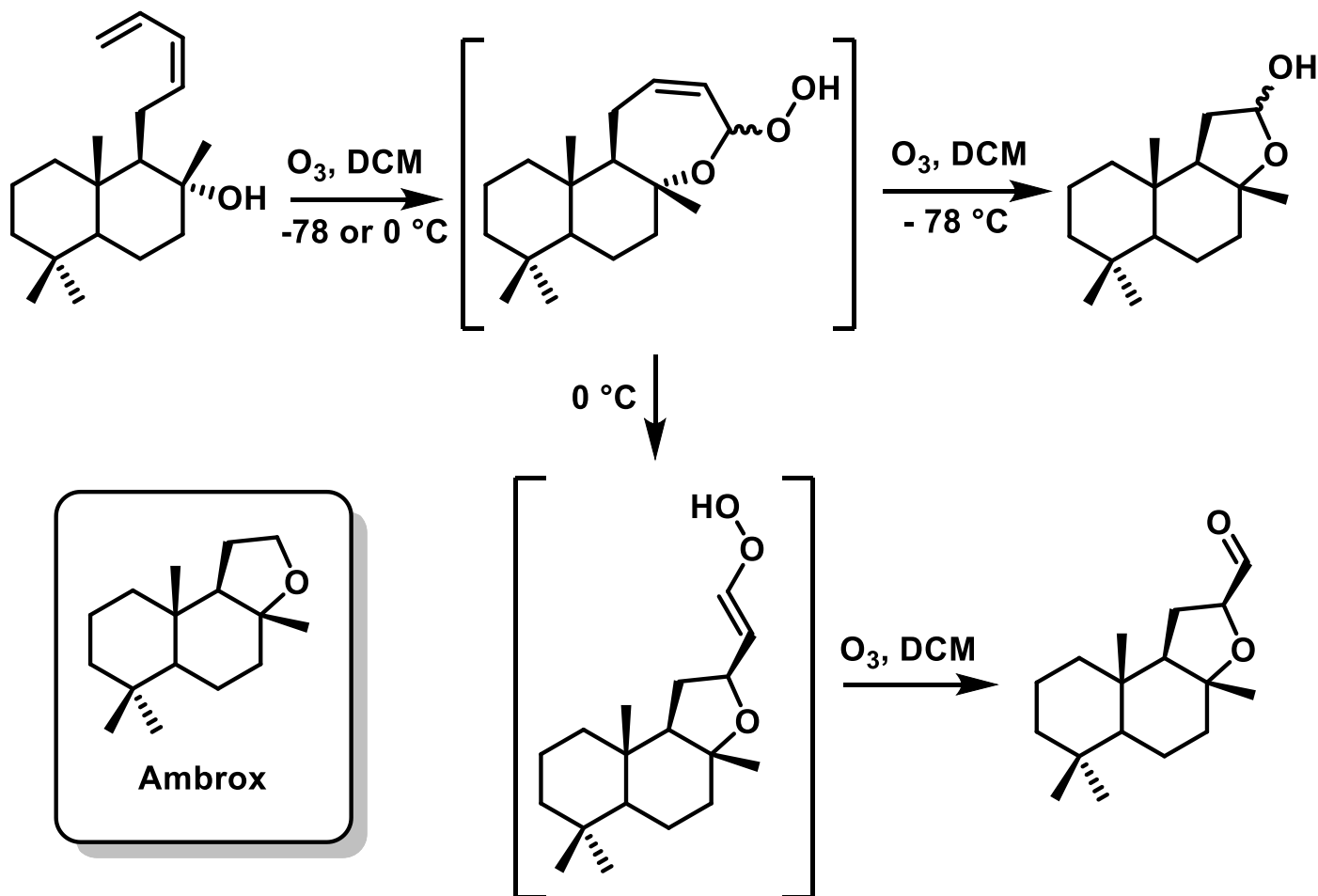
Dealkenylative Alkynylation



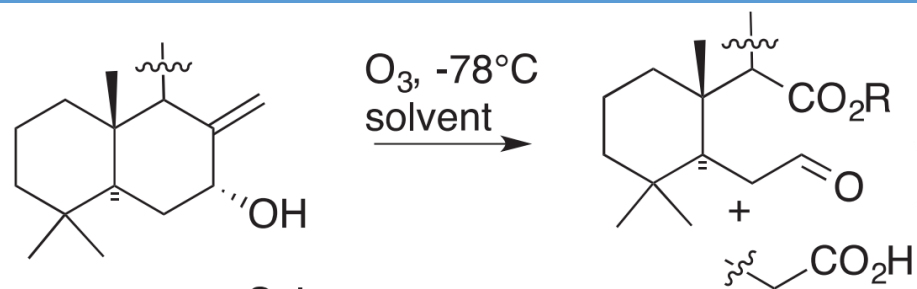
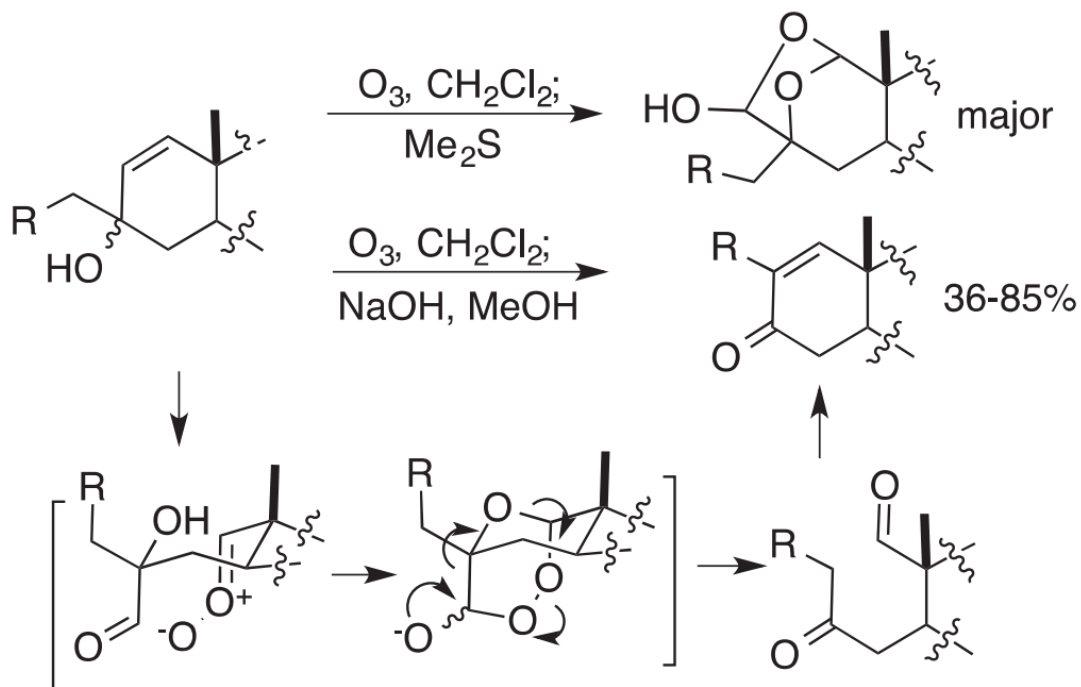
Hydroperoxyl Ketal from Furan



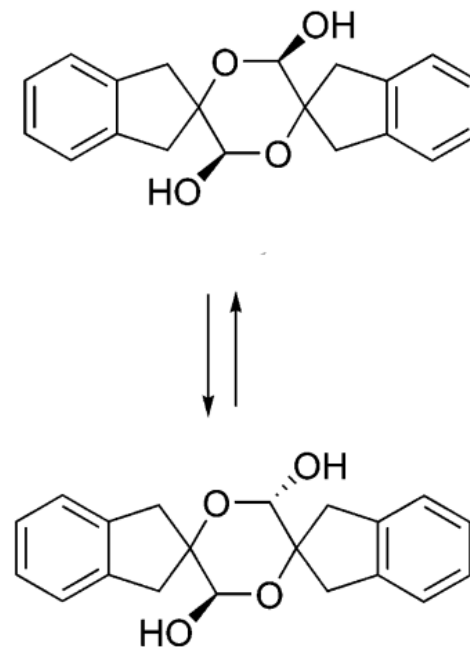
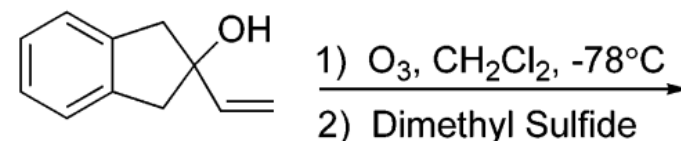
Diene: 1,2 to 1,4 Rearrangement



Allyl Alcohol: Fragmentation

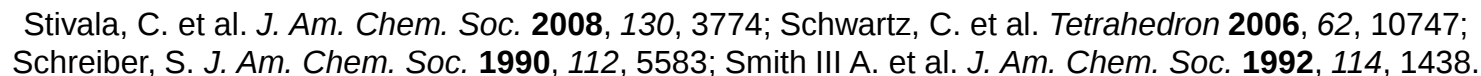
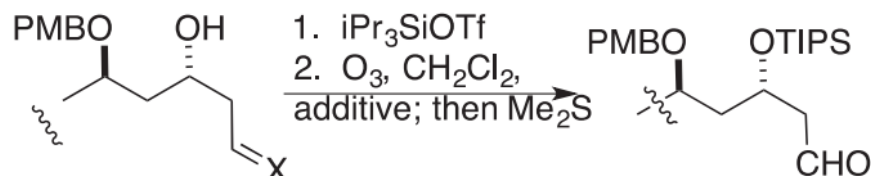


Solvent	Major product
CH_2Cl_2	$\text{R} = \text{H}$
MeOH	$\text{R} = \text{Me}$

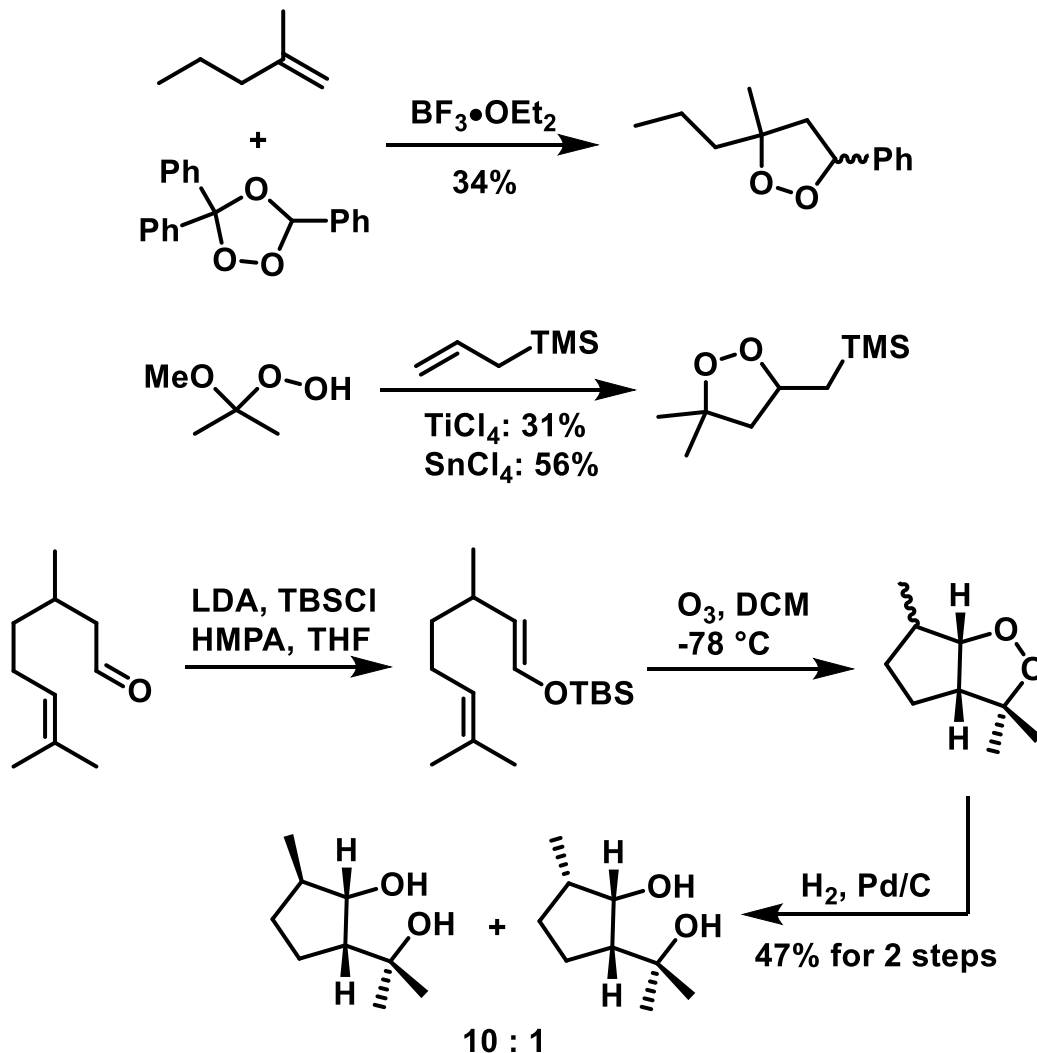


Ragan, J. et al. *Org. Process. Res. Dev.* **2003**, 7, 155;
 Everest, D. et al. *Aust. J. Chem.* **1988**, 41, 1025;
 DeNinno, M. et al. *J. Am. Chem. Soc.* **1995**, 117, 9927.

- Avoid peroxide accumulation
- Inhibit side reactions of CIs



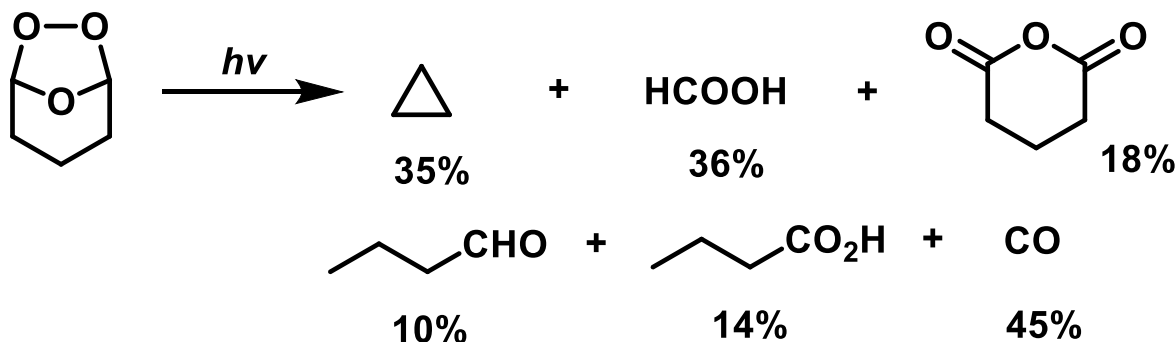
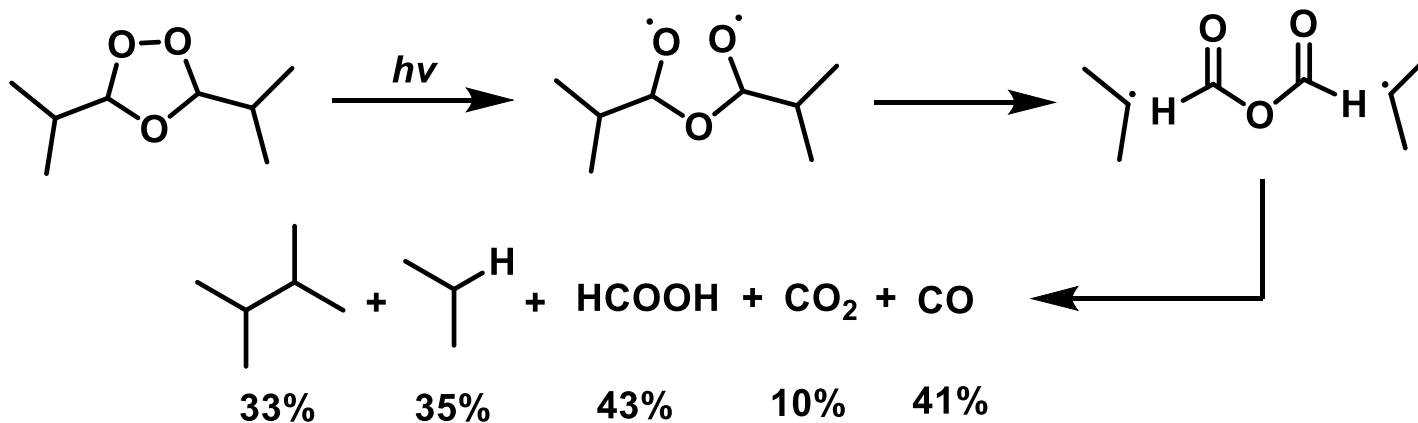
Criegee Intermediate as 1,3-Dipole



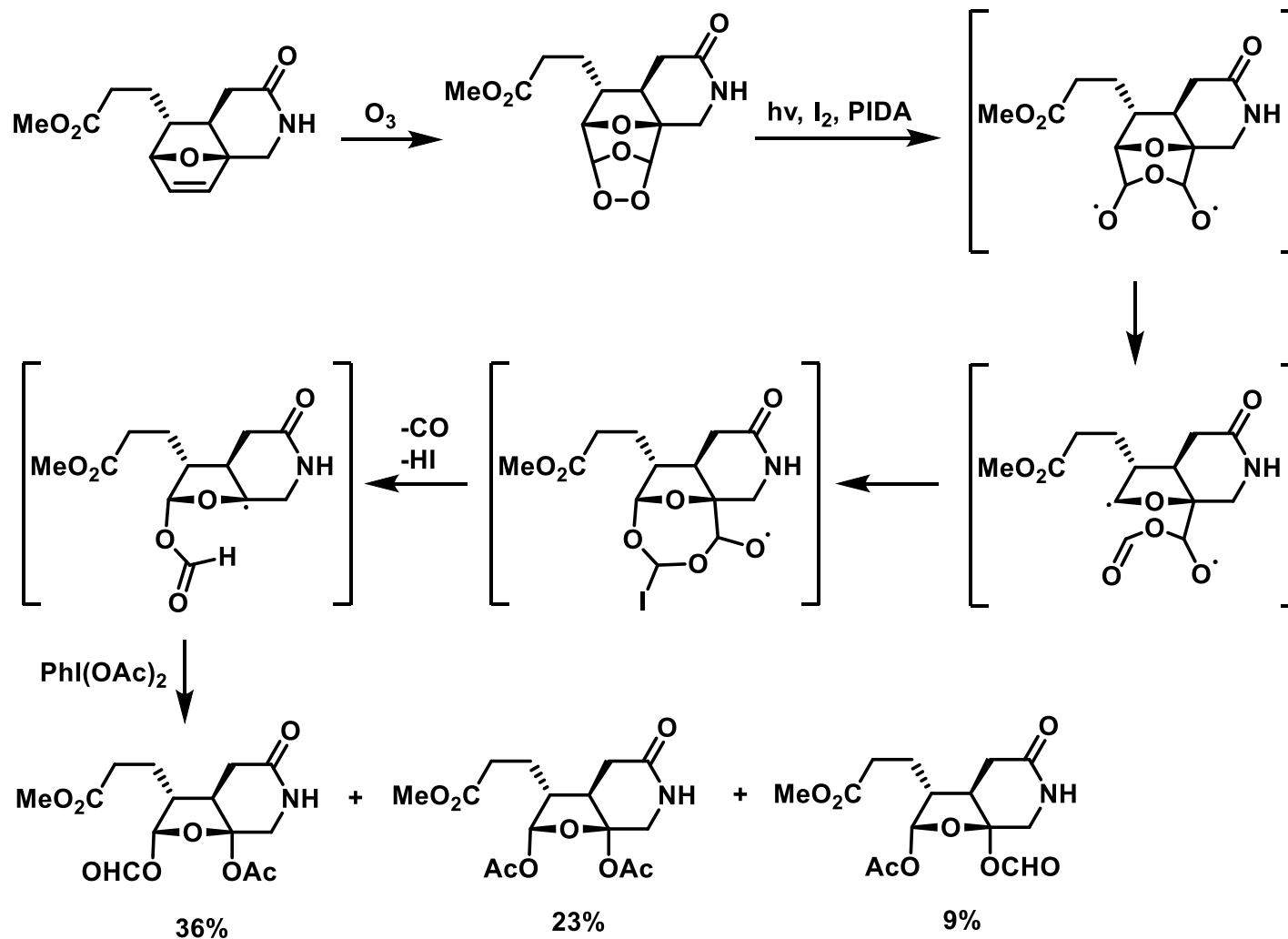
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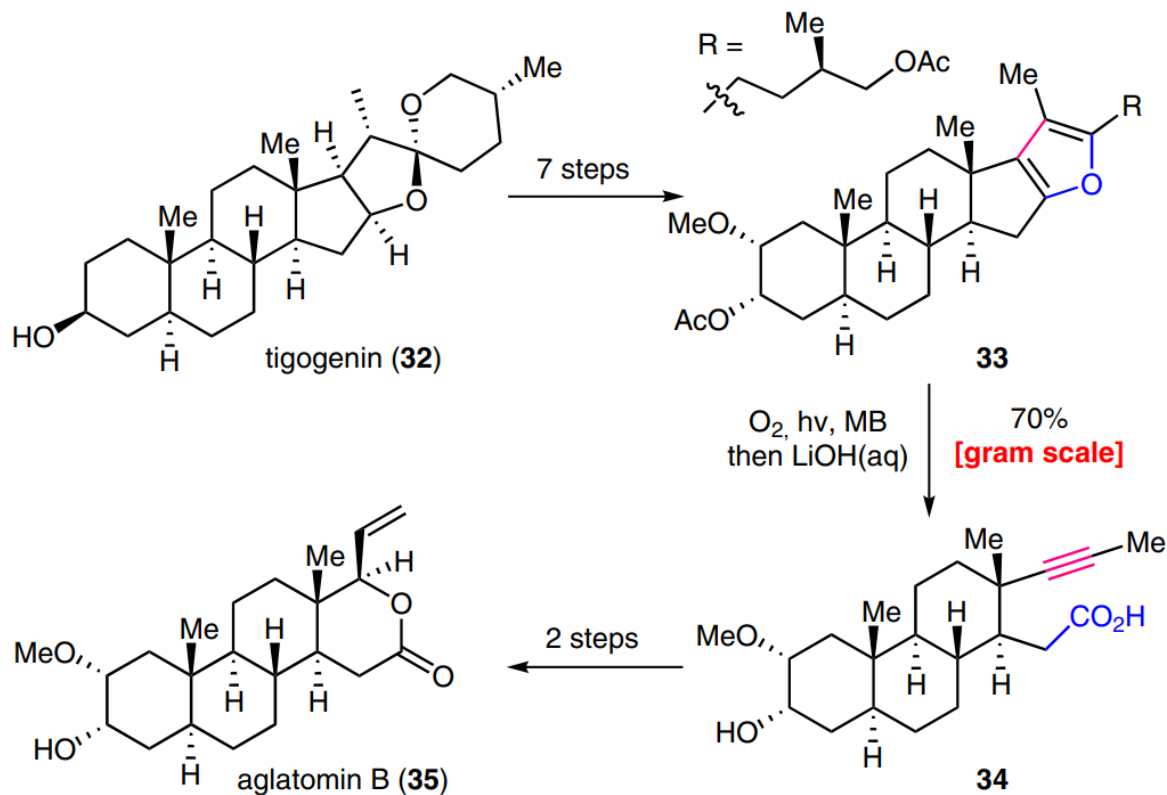
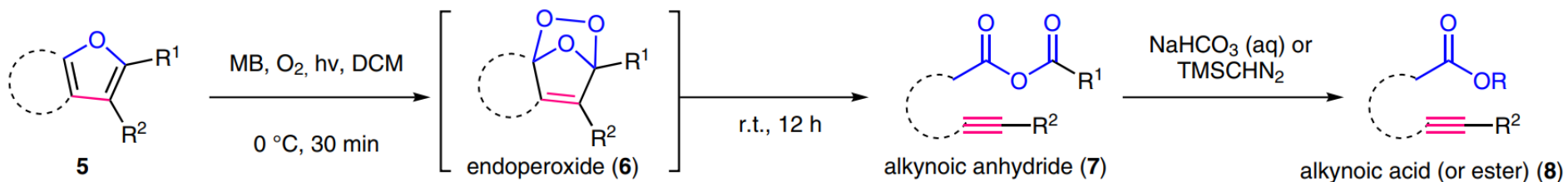
Pyrolysis or Photolysis: Story Reaction



Diradical Capture: PIDA



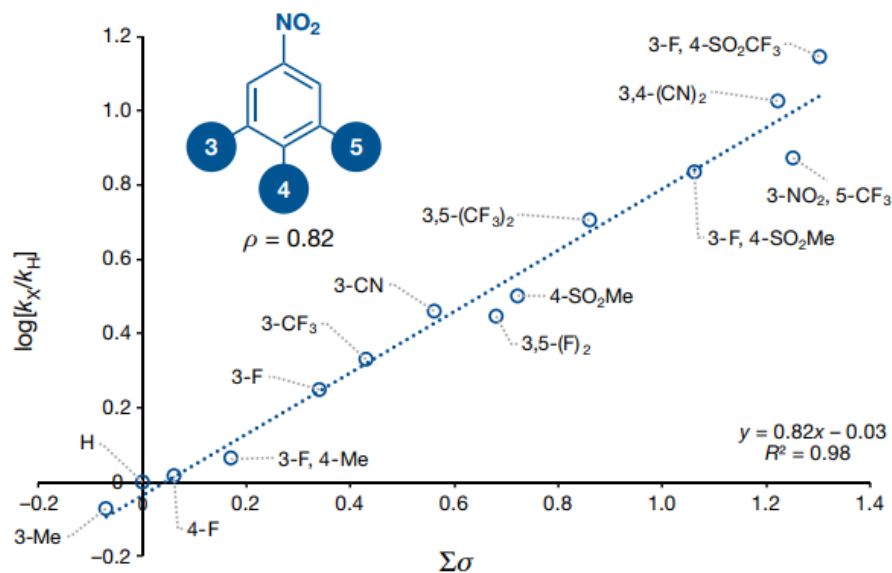
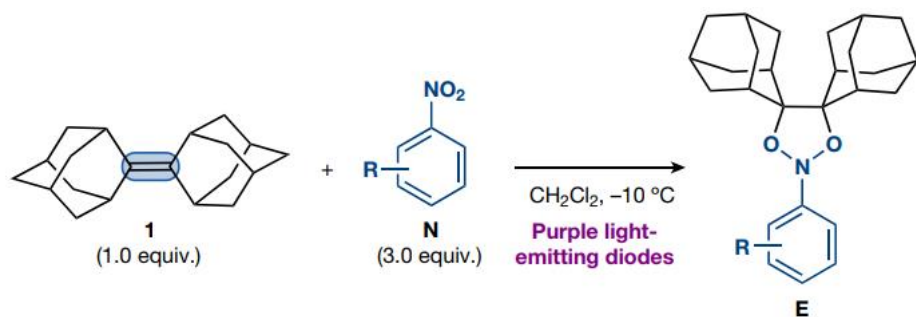
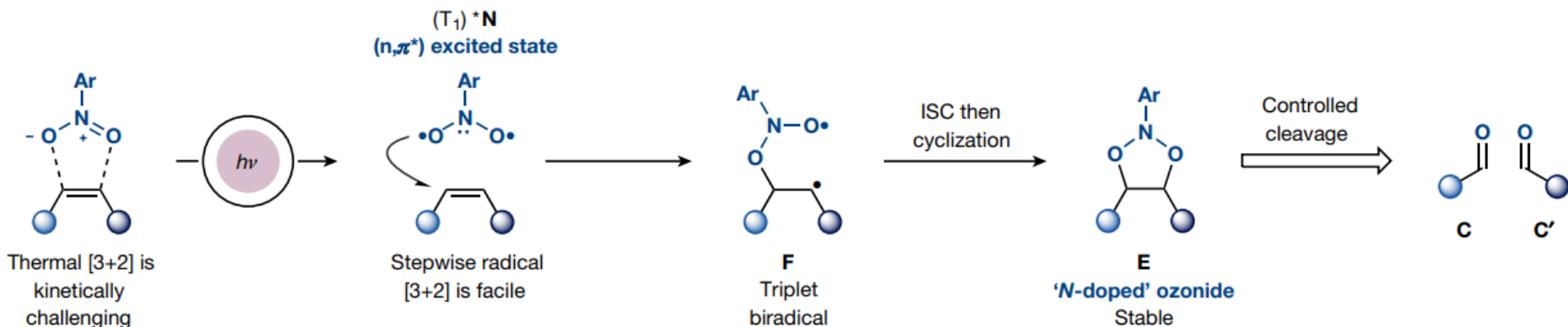
Alkyne-Forming Furan Fragmentation



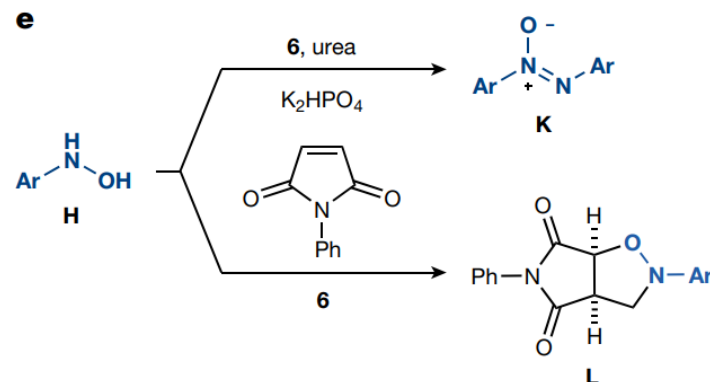
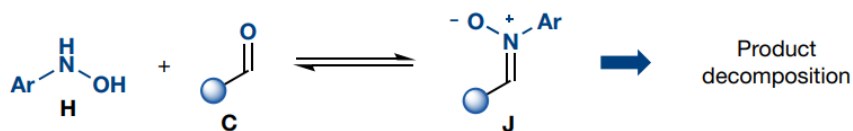
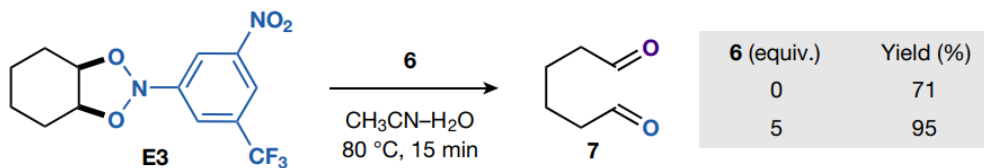
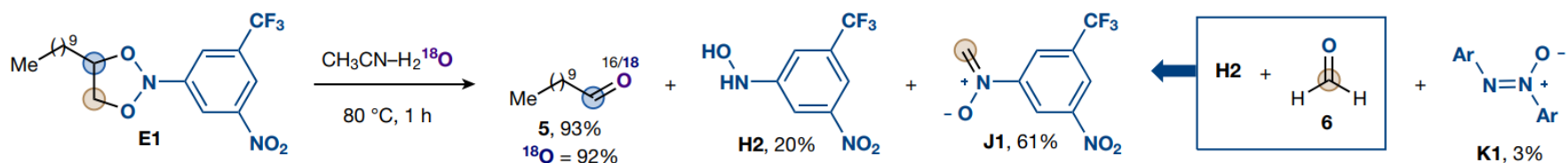
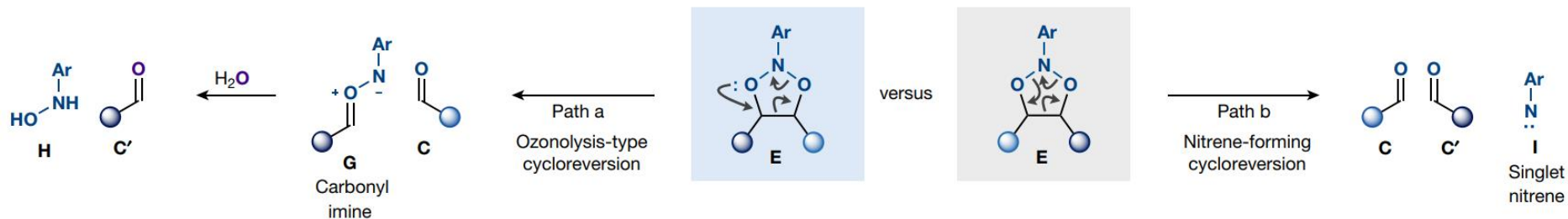
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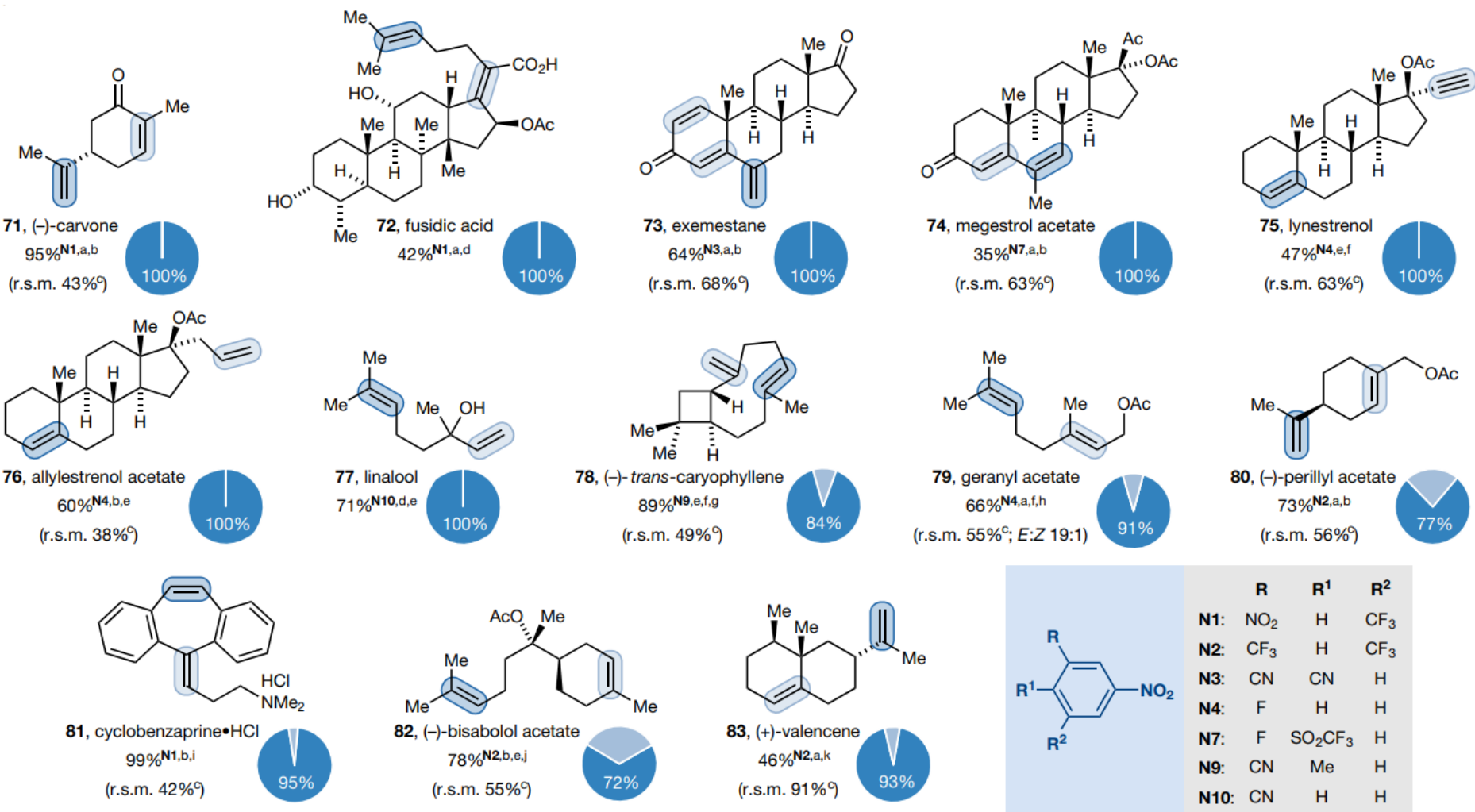
Photoexcited nitroarenes *Ozonolysis*



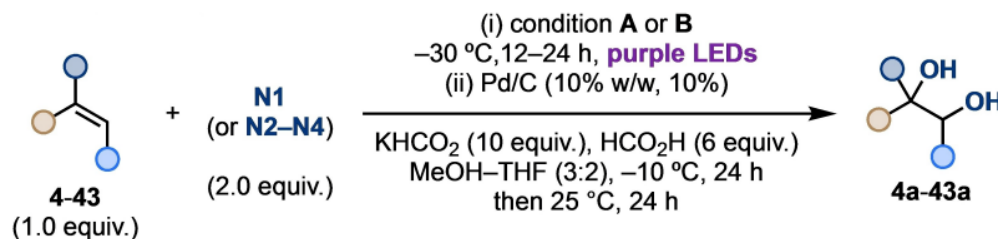
Ozonolysis-Type Fragmentation



Selective Olefin Oxidation Cleavage

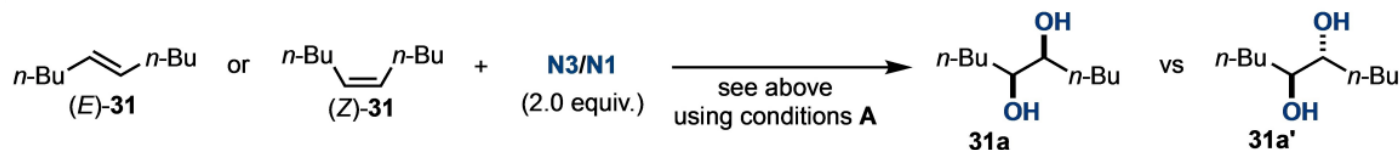


Dihydroxylation Using Nitroarenes

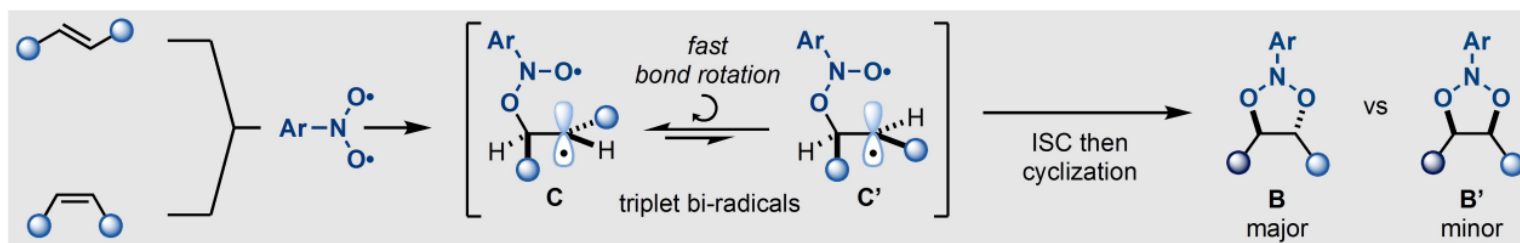


conditions	
A	EtOAc (0.17–0.50 M)
B	DCE (0.17–0.50 M)
	HFIP (4–5 equiv.)

	R	R ¹	R ²
N1	NO ₂	H	CF ₃
N2	F	SO ₂ CF ₃	H
N3	CF ₃	H	CF ₃
N4	F	H	H



entry	31	N	yield (%)	31a:31a'
1	<i>E</i>	N3	75	3.4:1
2	<i>Z</i>	N3	66	2.7:1
3	<i>E</i>	N1	68	3.3:1
4	<i>Z</i>	N1	64	2.5:1



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