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Large-scale aligned crystalline CH₃NH₃PbI₃ perovskite array films⁺

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Owing to the lattice match, aligned $CH_3NH_3PbI_3$ perovskite array films composed of $CH_3NH_3PbI_3$ microribbons with the length at the submillimeter scale were fabricated on ST cut quartz which showed a fast photo-response. This method provides a simple and flexible approach to grow oriented crystalline $CH_3NH_3PbI_3$ arrays at the millimeter scale.

Organic-inorganic hybrid lead halide perovskite (CH₃NH₃PbX₃, X = Cl, Br, I) based solar cells pioneered by Kojima *et al.* have obtained a significant improvement in power conversion efficiency from 3.8% to 20.1% in the past five years.¹⁻²² It is well known that the diffusion length of perovskite layers, which is closely related to the film crystallinity and morphology, plays a key role in the performances of the device. Numerous efforts have been made to enhance the photovoltaic performance by improving the crystallinity of CH₃NH₃PbX₃ films and therefore increasing the diffusion length.^{2,13-17,20,23} Horváth et al. first prepared lead iodide perovskite (CH₃NH₃PbI₃) nanowires with a length up to 10 µm.24 Shi et al. obtained crack-free MAPbX3 single crystals exceeding 100 mm³ in volume and observed low trap-state density and large carrier diffusion lengths above 10 µm.23 Nie et al. reported solution-processed perovskite solar cells with millimeter-scale grains and recorded efficiencies approaching 18% with little cell-to-cell variability.20

Besides the crystallinity, it is known that the organization or the arrangements of the materials is also an important issue. For example, oriented arrays of one-dimensional single crystalline ZnO and TiO_2 , compared to random polycrystalline nanostructures showed longer electron diffusion length and faster charge transport, which provides a solution to further improve the performances of solar devices.^{25,26} Yet, until now, there is no report on the preparation of well-aligned lead halide perovskite crystals.

Epitaxy has shown to be a powerful strategy to prepare ordered crystalline nanostructures of a wide variety of materials such as giant magnetoresistance nanostructures, semiconducting thin films, and graphene.^{27–31} Lattice match between the substrate and the target material is essential in epitaxial growth. With a suitable substrate, we may be able to epitaxially grow ordered CH₃NH₃PbX₃ crystals.

We find that the surface of ST cut quartz, which has been widely used to guide the growth of horizontally aligned singlewalled carbon nanotube (SWNT) arrays,32-35 shows similar structural characteristics to tetragonal CH₂NH₂PbI₂ as shown in Scheme 1. Along the [1 0 0] direction on the surface of ST cut quartz, which is just the direction SWNT oriented, there is a structural period with a scale of 19.65 Å (l_a) . The lattice spacing of (0 0 1) planes of CH₃NH₃PbI₃ is 12.68 Å (d_{001}).¹ It can be found that $2l_a \approx 3d_{001}$ with a mismatch of 3.2%. (1 0 1) plane in CH₃NH₃PbI₃, whose dihedral angle with the (0 0 1) plane is 55.0°, present a lattice spacing of 7.30 Å (d_{101}). In the surface lattice of ST cut quartz, along the direction of 54.5° apart from [1 0 0], a period of 16.91 Å (l_b) exists. It can also be found that $2l_{\rm b} \approx 5d_{101}$ with a mismatch of 7.0%. This length adaption is not perfect but still acceptable. However, the angle mismatch is only 0.9%. Hence, ST cut quartz might be possible to act as the substrate for the epitaxial growth of crystallized CH₃NH₃PbI₃ structures relying on the lattice match between the two materials. Also, the surface structure of X, Y, and Z cut quartz was also investigated as shown in Fig. S1[†] and the periodic lengths are listed in Table S1.† They do not show a good structural match to the CH₃NH₃PbI₃ crystal.

We tried to use the slip-coating²⁴ method to grow CH_3NH_3 -PbI₃ crystals on the ST cut quartz substrate with DMF (*N*,*N*dimethyl formamide) solution and a glass slide as the slip. As

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Scheme 1 The structure matching between $CH_3NH_3PbI_3$ and ST cut quartz. (a) Two unit cells of $CH_3NH_3PbI_3$; (b) d_{001} and d_{101} shown on the (0 1 0) plane of $CH_3NH_3PbI_3$; (c) periodical structure on the surface of ST cut quartz.

shown in Fig. 1a and b, aligned microribbon arrays with the length of hundreds of microns were grown on the surface of ST cut quartz. All peaks in the X-ray diffraction (XRD) pattern of the

sample can be readily assigned to tetragonal CH₃NH₃PbI₃ and the quartz substrate as shown in Fig. S2.[†] The intensive peaks appearing at 2 θ of 13.9°, 14.3°, 28.0°, and 28.6° correspond to diffractions of (0 0 2), (1 1 0), (0 0 4), and (2 2 0) planes of CH₃NH₃PbI₃, respectively.¹ This indicates that well crystallized perovskite with good phase purity was formed on quartz. The high resolution transmission electron microscopy (HRTEM) image in Fig. 1d shows fringes perpendicular to the longitudinal direction of the microribbon with the inter-distance of 3.2 Å, which is in good accordance with d_{004} (3.17 Å) of CH₃-NH₃PbI₃. The Fourier transform pattern of the HRTEM image can also be well ascribed to tetragonal CH₃NH₃PbI₃. Therefore the microribbons extend along the [0 0 1] direction or *c* axis of the CH₃NH₃PbI₃ crystal.

We also performed slip-coating under the same conditions on X, Y, and Z cut quartz, respectively. On these quartz substrates, $CH_3NH_3PbI_3$ crystals were randomly oriented (Fig. 2b–d). But on ST cut quartz, no matter what direction we moved the slipping slide, the $CH_3NH_3PbI_3$ microribbons were always oriented along the same direction, *i.e.* 45° to the margins of the quartz substrates as shown in Fig. 3. This is the same direction as the aligned SWNT arrays on ST cut quartz and we know it is the [1 0 0] direction of quartz.^{32,35} All these facts reveal that the surface lattice of ST cut quartz is responsible for the formation of aligned $CH_3NH_3PbI_3$ ribbons. Taking the HRTEM characterization in Fig. 1 into account, it is obvious that the [0 0 1] direction of $CH_3NH_3PbI_3$ parallels the [1 0 0] direction of quartz. This is in good accordance to what we already predicted in Scheme 1.



Fig. 1 SEM and TEM characterization of CH₃NH₃PbI₃ microribbons formed on ST cut quartz. (a and b) Low (a) and high magnification (b) SEM images. The inset shows the TEM image of a single perovskite microribbon. The bar represents 2 μ m. (c and d) TEM (c) and the HRTEM (d) image of the site pointed by the arrow in (c). The corresponding FFT pattern is inserted in (d).



Fig. 2 SEM images of $CH_3NH_3PbI_3$ performed on ST (a), X (b), Y (c), and Z (d) cut quartz and the tapping-mode AFM topographic image (e) as well as the 3D height image (f) of the aligned $CH_3NH_3PbI_3$ arrays at different magnifications.



Fig. 3 SEM images of $CH_3NH_3PbI_3$ microribbons fabricated in different slip-coating directions.

We carried out two-dimensional wide angle X-ray diffraction (WAXD) measurements in the 2θ range of 0 to 30° to further verify the structural correlation between CH₃NH₃PbI₃ microribbons and the ST cut quartz. In the two-dimensional WAXD images, 2θ is represented by the distance to the center. Blank ST cut quartz only shows one diffraction spot at a 2θ of ~26.6°



Fig. 4 SEM images, 2D WAXD pattern and XRD patterns of the substrate with aligned MAPbl₃ microribbon arrays (a and c) and CH₃-NH₃Pbl₃ microribbons with disordered orientations (b and d).

(Fig. S3[†]), which was assigned to (1 0 1) planes of ST cut quartz (JCPDS no. 46-1045). This spot also appears in the diffraction patterns of ST cut quartz substrates with CH₃NH₃PbI₃ microribbons (Fig. 4c and d). CH₃NH₃PbI₃ microribbon arrays presents a strong diffraction spot at $\sim 13.9^{\circ}$ (Fig. 4c); while disordered CH₃NH₃PbI₃ microribbons only show a weak ring at the same diffraction angle (Fig. 4d), which corresponds to $(0\ 0\ 2)$ planes of CH₃NH₃PbI₃. This result indicates the good alignment of CH₃NH₃PbI₃ microribbon arrays obtained by the slip-coating method. In WAXD patterns, the angle between the (0 0 2) plane of CH₃NH₃PbI₃ and the (1 0 1) plane of ST cut quartz is about 43.0° . We know that the angle between $(1\ 0\ 1)$ planes and $(1\ 0\ 0)$ planes of ST cut quartz is $\sim 42.3^{\circ}$. Therefore, the *c* axis of CH₃NH₃PbI₃ microribbons is parallel to the [1 0 0] direction of quartz. The WAXD measurements reveal the orientation of the CH₃NH₃PbI₃ crystal on ST cut quartz and provides further evidence for the epitaxial growth mechanism.

The thickness of the liquid films on the quartz substrate can be varied by changing the pressure exerted onto the slipping slide, and subsequently the thickness of the $CH_3NH_3PbI_3$ microribbons can be easily adjusted. As shown in Fig. S4,† the average heights have been modulated from ~400 nm to ~2 µm, though some sophisticated techniques are needed to realize a precise control. Fig. 2e and f show the typical AFM images of the $CH_3NH_3PbI_3$ microribbon arrays with a thickness of ~700 nm. The widths of the ribbons are around hundreds of nanometers to several microns.

Because the CH₃NH₃PbI₃ microribbons were formed on quartz, the samples are ready for direct detection of the UV-Vis-NIR absorption. The microribbons show intensive and broad absorption from UV to the near IR region until ~800 nm as shown in Fig. S6.† The bandgap was estimated to be ~1.6 eV, which is very similar to the reported values.^{1,4}

Au electrodes with a thickness of 60 nm and 8 nm of Cr buffer layer were evaporated directly onto the quartz substrates with CH₃NH₃PbI₃ microribbons to fabricate devices and their photo-response performances were studied (Fig. 5). The



Fig. 5 The devices of $CH_3NH_3PbI_3$ microribbon arrays with Au as electrodes and the photoresponse performance. (a and b) Images of the devices with the electrodes at low and high amplifications; (c and d) I-V (c) and I-T (d) characteristics of devices.

channel lengths between two electrodes are about 40 μ m and the side lengths of the square Au electrodes are ~200 μ m. We measured the *I*-*T* curves at an ON/OFF interval of ~10 s. The response time of the photocurrent was observed to be ~0.1 s, which actually already reaches the limitation of our detector. The photocurrent of devices built on aligned microribbons is normally 5 times higher than the device of disordered CH₃-NH₃PbI₃ microribbons (Fig. S5†). In addition, a sample annealed at 80 °C for 2 hours was measured for comparison. No obvious difference in photoresponse performance was observed. This may indicate that MAPbI₃ microribbons formed by epitaxial growth are already with considerable crystallinity.

Conclusions

In summary, aligned CH₃NH₃PbI₃ array films with high-quality crystallization have been fabricated on ST cut quartz via a slipcoating method. The surface lattice of ST cut quartz acts as the template to guide the epitaxial growth of CH₃NH₃PbI₃ microribbon arrays. The matching between the lattice spacing of (0 0 1) planes of CH₃NH₃PbI₃ and the intervals along the [1 0 0] direction of ST cut quartz leads to the oriented growth of CH₃-NH₃PbI₃ microcrystals along the [1 0 0] direction of ST cut quartz, resulting in CH₃NH₃PbI₃ microribbons extended along the *c* axis. These ribbons are in the length of the sub-millimeter scale, width of microns, and thickness of hundreds of nanometers to several microns depending on the preparation conditions. The elimination of grain boundaries is always desired for perovskite-based devices. The epitaxial growth on ST cut quartz provides a strategy to produce large scale aligned microribbon perovskite films with high crystallinity, which may bring about improved performance. In addition, the alignment of the microribbons should lead to heterogeneous performance along different directions, which can also be made use of in certain applications.

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