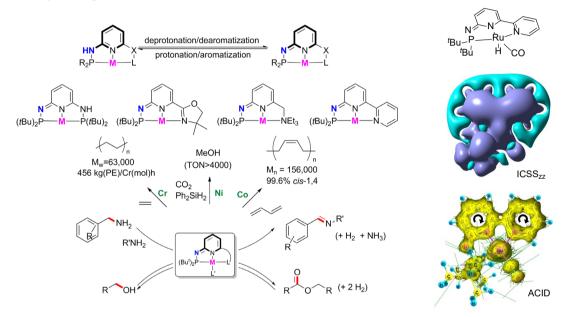
PN³(P)Pincer Complexes: Cooperative Catalysis and Beyond

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Pincer transition metal complexes have versatile reactivities to catalyze many organic transformations and activate strong chemical bonds. In particular, complexes with ligands derived from tridentate pyridine-based framework exhibit interesting reactivities. We have developed a novel platform of pincer-type $PN^3(P)$ -ligands which are capable of interacting with the substrates during the reaction. We have witnessed that the seemingly small change by replacing the CH_2 spacer in the pyridine-based pincer complex with an NH group has dramatically influenced the thermodynamic and kinetic properties and in some cases the catalytic behaviors of the corresponding metal complexes. Furthermore, unprecedented ligand-centered reactivities (LCRs) were demonstrated. The σ nucleophilicity of the N atom of the iminic arm was significantly enhanced to reach N-heterocyclic carbene-like nucleophilicity as a catalyst and a ligand. The novel LCR may open a new direction for the catalyst design.



Key References

(1) Li, H.; Gonçalves, T. P.; Lupp, D.; Huang, K.-W. "PN³(P)-Pincer Complexes: Cooperative Catalysis and Beyond" ACS Catal. 2019, 9, 1619-1629. (2) Gonçalves, T. P.; Dutta, I.; Huang, K.-W. "Aromaticity in catalysis: metal ligand cooperation via ligand dearomatization and rearomatization" Chem. Commun. 2021, 57, 3070-3082. (3) US Patent No. 8,598,351 (2013); Title: Phospho-Amino Pincer-Type Ligands and Catalytic Metal Complexes Thereof. (4) Li, H.; Zheng, B.; Huang, K.-W. "A new class of PN³-pincer ligands for metal-ligand cooperative catalysis" Coord. Chem. Rev. 2015, 293-294, 116-138. (5) Gonçalves, T. P.; Huang, K.-W. "Metal-Ligand Cooperative Reactivity in the (pseudo)-Dearomatized PN^x(P) Systems: the Influence of the Zwitterionic Form in Dearomatized Pincer Complexes "J. Am. Chem. Soc. 2017, 139, 13442-13449. (6) Eppinger, J.; Huang, K.-W. "Formic Acid as a Hydrogen Energy Carrier" ACS Energy Lett. 2017, 2, 188-195. (7) Zhang, Y.; Chen, X.; Zheng, B.; Guo, X.; Pan, Y.; Chen, H.; Li, H.; Min, S.; Guan, C.; Huang, K.-W.; Zheng, J. "Structural Analysis of Transient Reaction Intermediate in Formic Acid Dehydrogenation Catalysis Using Two-Dimensional IR Spectroscopy" Proc. Natl. Acad. Sci. U.S.A. 2018, 115, 12395-12400. (8) Li, H.; Gonçalves, T. P.; Hu, J. Zhao, Q.; Gong, D.; Lai, Z.; Wang, Z.; Zheng, J.; Huang, K.-W. "A Pseudo-Dearomatized PN3P*Ni-H Complex as a New Ligand and σ-Nucleophilic Catalyst" J. Org. Chem. 2018, 83, 14969-14977. (9) Wang, X.; Ang, E. P. L.; Guan, C.; Zhang, W.; Wu, W.; Liu, P.; Zheng, N.; Zhang, D.; Lopatin, S.; Lai, Z.; Huang, K.-W. "Single-site ruthenium pincer complex knitted in porous organic polymers for green dehydrogenation of formic acid in aqueous medium" ChemSusChem 2018, 11, 3591-3598. (10) Chen, T.; Li, H.; Qu, S.; Zheng, B.; Lai, Z.-P.; Wang, Z.-X.; Huang, K.-W. "Hydrogenation of Esters Catalyzed by Ruthenium PN3-Pincer Complexes Containing an Aminophosphine Arm" Organometallics 2014, 33, 4152-4155. (11) Qu, S.; Dang, Y.; Song, C.; Wen, M.; Huang, K.-W.; Wang, Z.-X. "Catalytic Mechanisms of Direct Pyrrole Synthesis via Dehydrogenative Coupling Mediated by PNP-Ir or PNN-Ru Pincer Complexes: Crucial Role of Proton-Transfer Shuttles in the PNP-Ir System" J. Am. Chem. Soc. 2014, 136, 4974-4991.